

**1st Africa/Middle East expert meeting and
workshop on the health impact of airborne dust**

2-5 Nov 2015, Amman

ground observation of airborne dust

Sergio Rodríguez

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AEMET, Spain

dust, aerosols and pollutants

in-situ observations

PM_{10} and $PM_{2.5}$ levels

PM_{10} and $PM_{2.5}$ composition

complementary observations

observation network

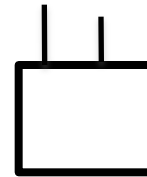


people live in cities and breath a cocktail dust + pollutants





people live in cities and breath a cocktail dust + pollutants



dust - air quality stations



parameters indicative of:

dust
ambient air quality

dust, aerosols and pollutants

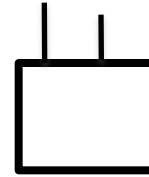
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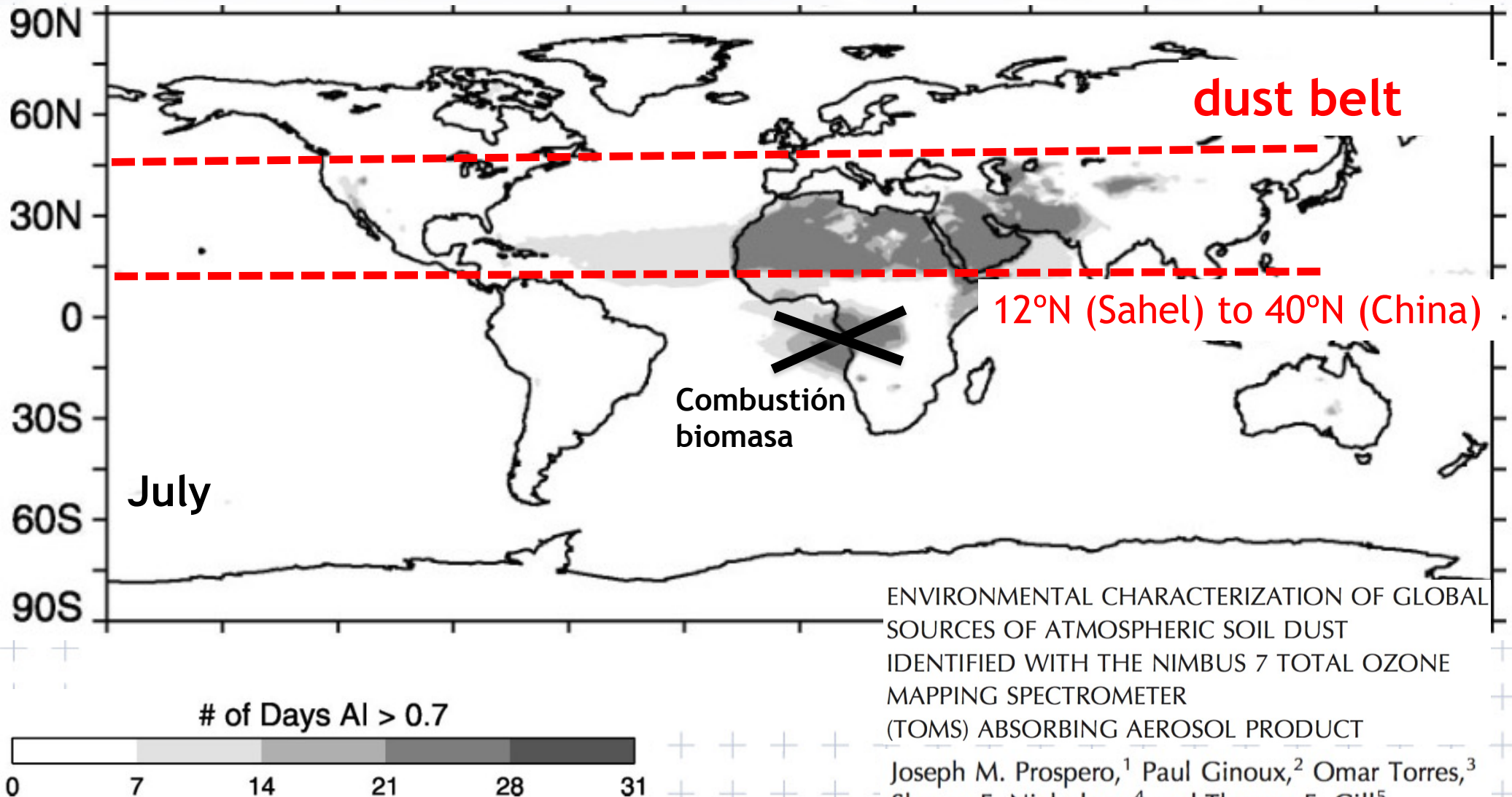


people live in cities and breath a cocktail dust + pollutants

what is dust?

UV absorbing aerosols - dust Satellite

What is dust?



ENVIRONMENTAL CHARACTERIZATION OF GLOBAL SOURCES OF ATMOSPHERIC SOIL DUST IDENTIFIED WITH THE NIMBUS 7 TOTAL OZONE MAPPING SPECTROMETER (TOMS) ABSORBING AEROSOL PRODUCT

Joseph M. Prospero,¹ Paul Ginoux,² Omar Torres,³ Sharon E. Nicholson,⁴ and Thomas E. Gill⁵

dust, aerosols and pollutants

chotts, sabkhas, wadis, salares

1. what is dust ?

There are several types of sources, but the mayor dust sources are associate with dry lakes/rivers beds

2. chemistry and mineralogy

clays, feldspars, oxides, evaporites

Si, Al, Ca, Fe, Mg, Na, Cl, Mn....

Table 6. Density and Real Index of Refraction of Minerals Found in Saharan Dust^a1. clays

illite	$\text{K}_{0.6}(\text{H}_3\text{O})_{0.4}\text{Al}_{1.3}\text{Mg}_{0.3}\text{Fe}_{0.1}\text{Si}_{3.5}\text{O}_{10}(\text{OH})_2 \cdot (\text{H}_2\text{O})$
kaolinite	$\text{Al}_2\text{Si}_2\text{O}_5(\text{OH})_4$
montmorillonite	$(\text{Na}, \text{Ca})_{0.5}(\text{Al}, \text{Mg}, \text{Fe})_4(\text{Si}, \text{Al})_8\text{O}_{20}(\text{OH})_4 \cdot n(\text{H}_2\text{O})$
smectite	$(\text{Na}, \text{Ca})\text{Al}_4(\text{Si}, \text{Al})_8\text{O}_{20}(\text{OH})_4 \cdot 2(\text{H}_2\text{O})$
chlorite	$\text{Na}_{0.5}(\text{Al}, \text{Mg})_6(\text{Si}, \text{Al})_8\text{O}_{18}(\text{OH})_{12} \cdot 5(\text{H}_2\text{O})$

2. evaporites

calcite	CaCO_3
dolomite	$\text{CaMg}(\text{CO}_3)_2$
gypsum	$\text{CaSO}_4 \cdot 2(\text{H}_2\text{O})$
anhydrite	CaSO_4
halite	NaCl

4. oxides

hematite	Fe_2O_3
goethite	$\text{FeO}(\text{OH})$
rutile	TiO_2

3. feldspars

microcline	KAlSi_3O_8	Var oligoclase	$(\text{Na}, \text{Ca})(\text{Si}, \text{Al})_4\text{O}_8$
		Var albite	$\text{NaAlSi}_3\text{O}_8$
		Var anorthite	$\text{CaAl}_2\text{Si}_2\text{O}_8$

Characterization of African dust transported to Puerto Rico by individual particle and size segregated bulk analysis

Elizabeth A. Reid,^{1,2,3} Jeffrey S. Reid,³ Michael M. Meier,⁴ Michael R. Dunlap,⁴ Steven S. Cliff,⁴ Aaron Broumas,⁴ Kevin Perry,⁵ and Hal Maring⁶

JOURNAL OF GEOPHYSICAL RESEARCH, VOL. 108, NO. D19, 8591, doi:10.1029/2002JD002935, 2003

dust, aerosols and pollutants

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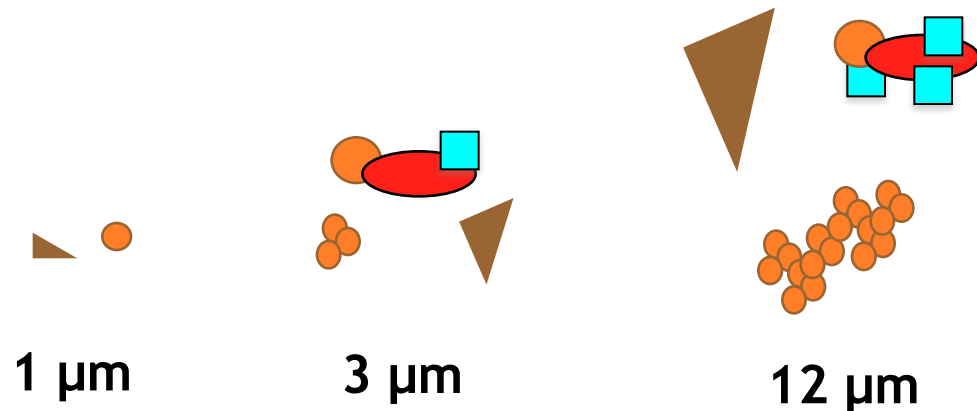
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3. Size and morphology

1 and 20 μm
agglomerates



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Chotts, Sabkhas



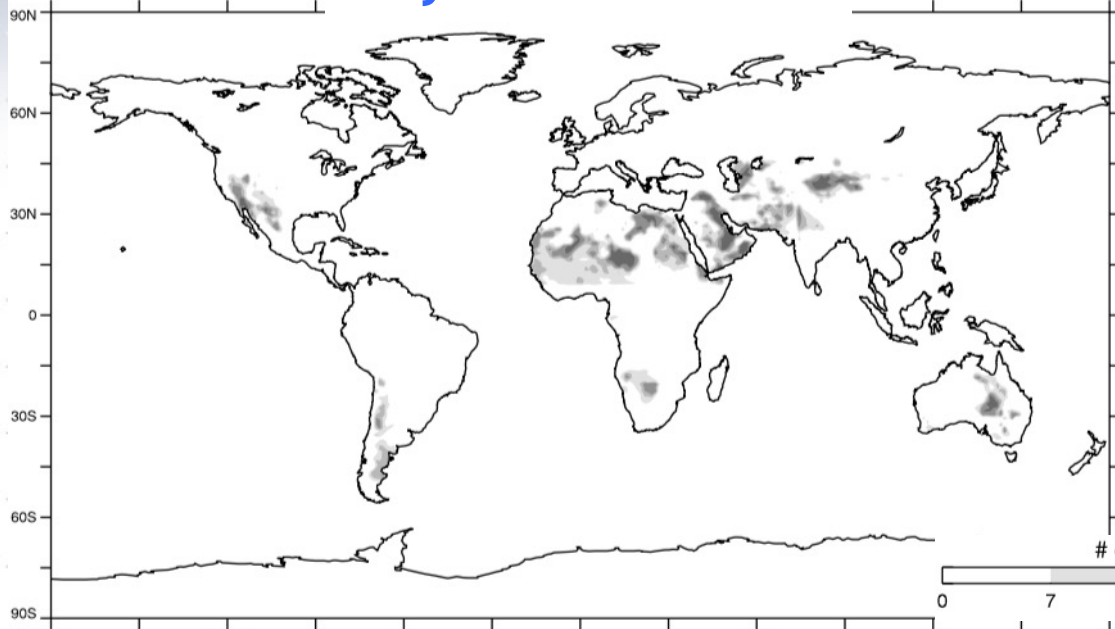
Dry lakes beds



wadis

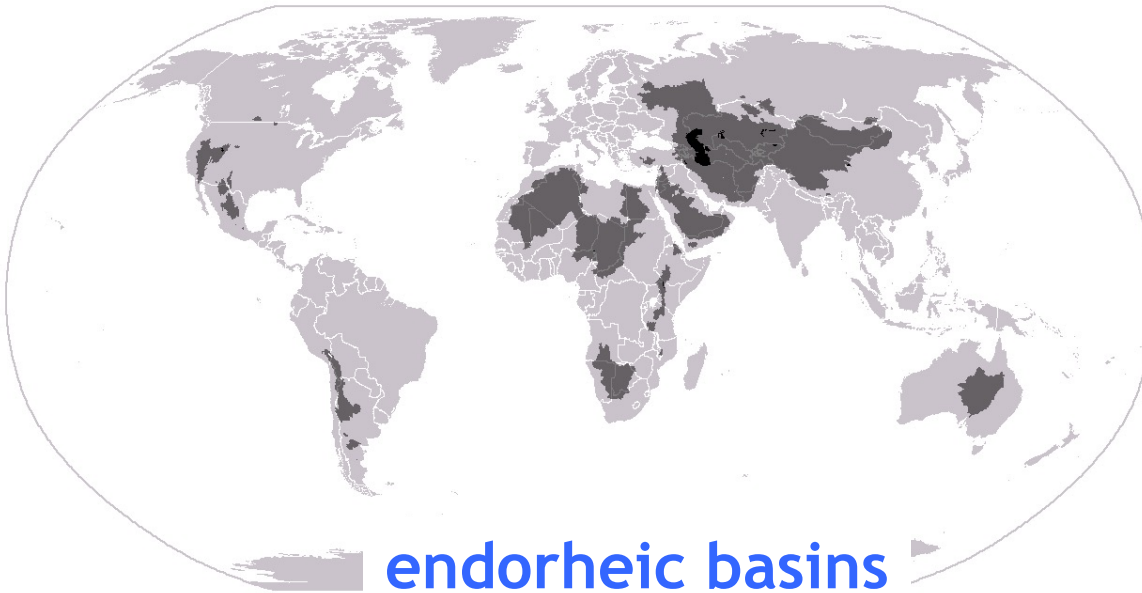


mayor dust sources



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IDENTIFIED WITH THE NIMBUS 7 TOTAL OZONE
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(TOMS) ABSORBING AEROSOL PRODUCT

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endorheic basins
no conection to sea

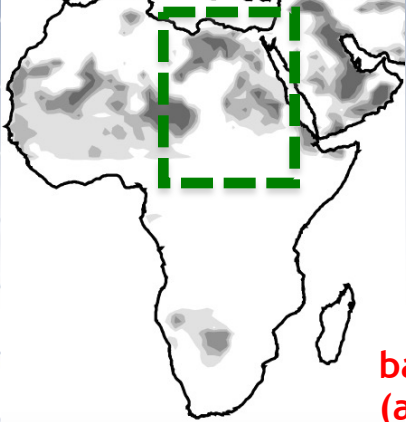


GOBIERNO
DE ESPAÑA

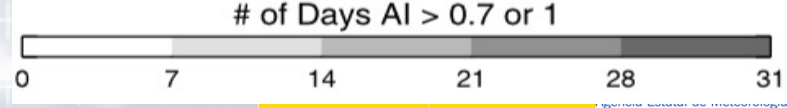
MINISTERIO
DE MEDIO AMBIENTE
Y MEDIO RURAL Y MARINO



Agencia Estatal de Meteorología



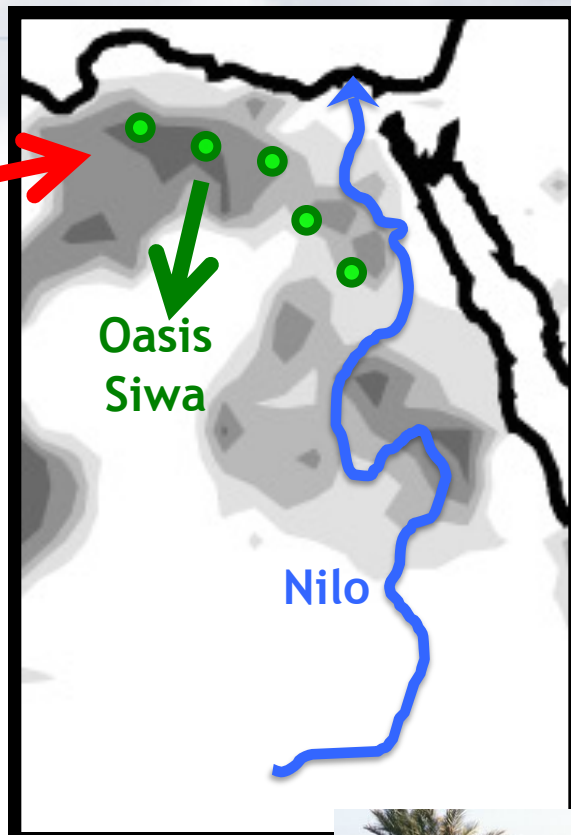
Detección satélite

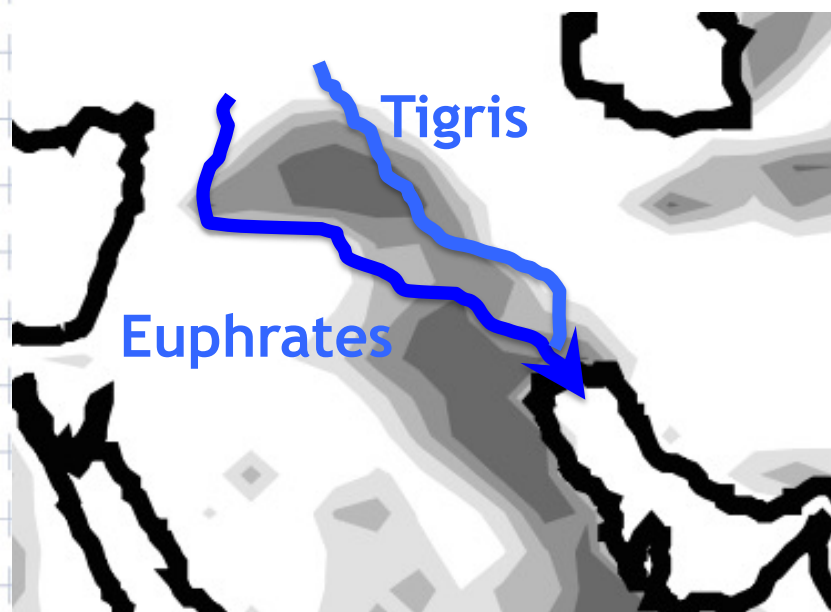
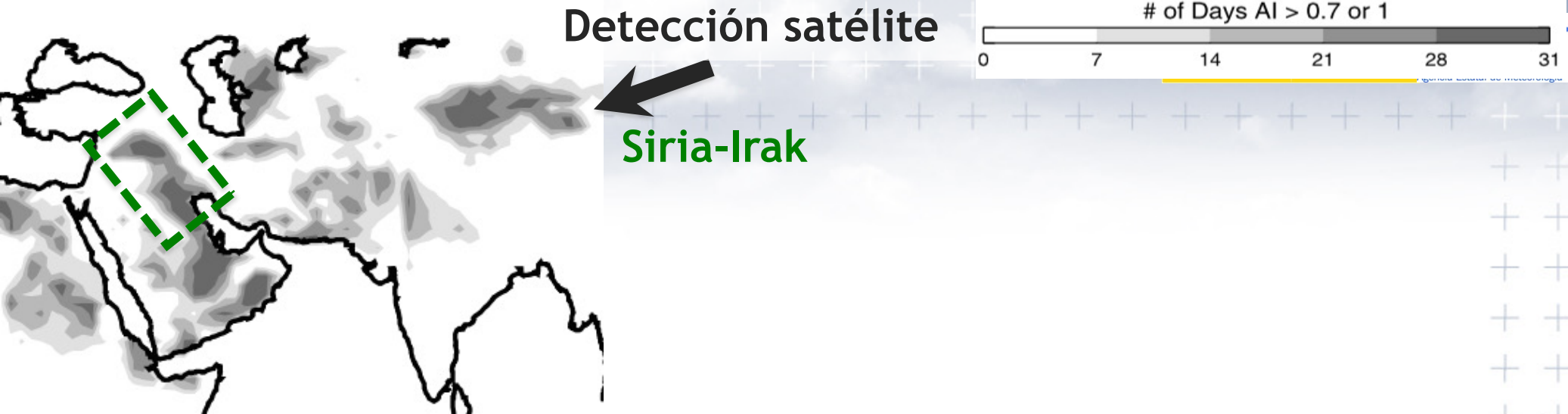


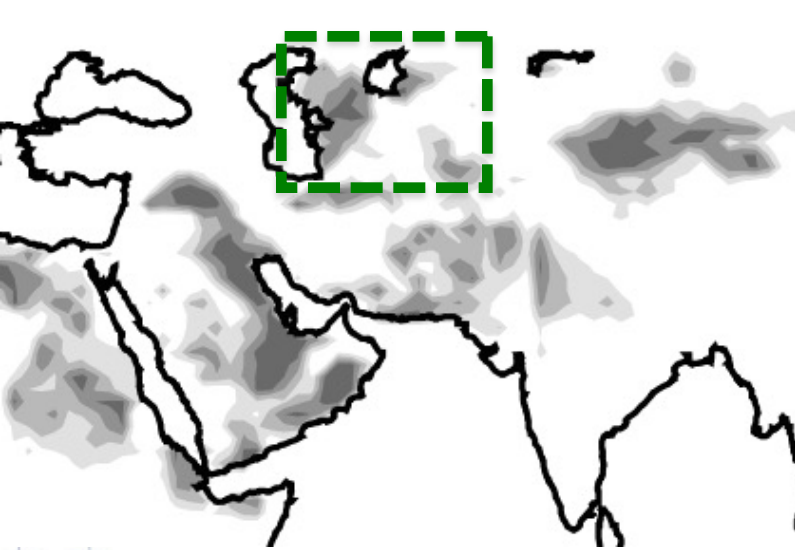
bajas topográficas
(altitudes -)

● lagos, oasis, cultivos
aguas subterráneas
antiguos ríos

Oasis Siwa
29°13'N, 25°31'E







During the 1960s, the Syr y Amu rivers were re- channelled for crop cultivation and the Aral Sea diminished increasing dust soruces

Caspian Kazakhstan Aral Sea
Sea



Aral Sea

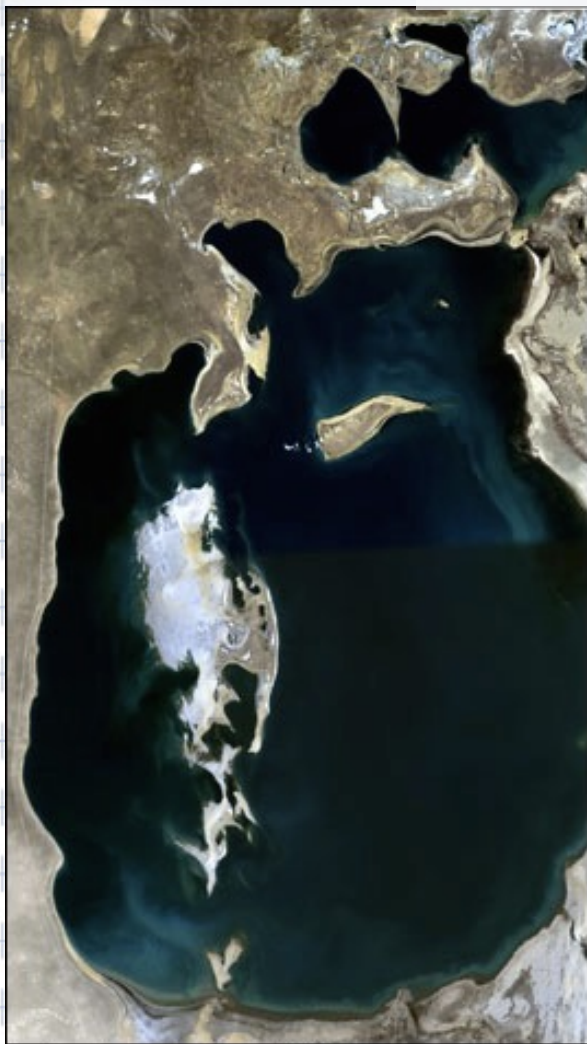


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Aemet
Agencia Estatal de Meteorología

1989



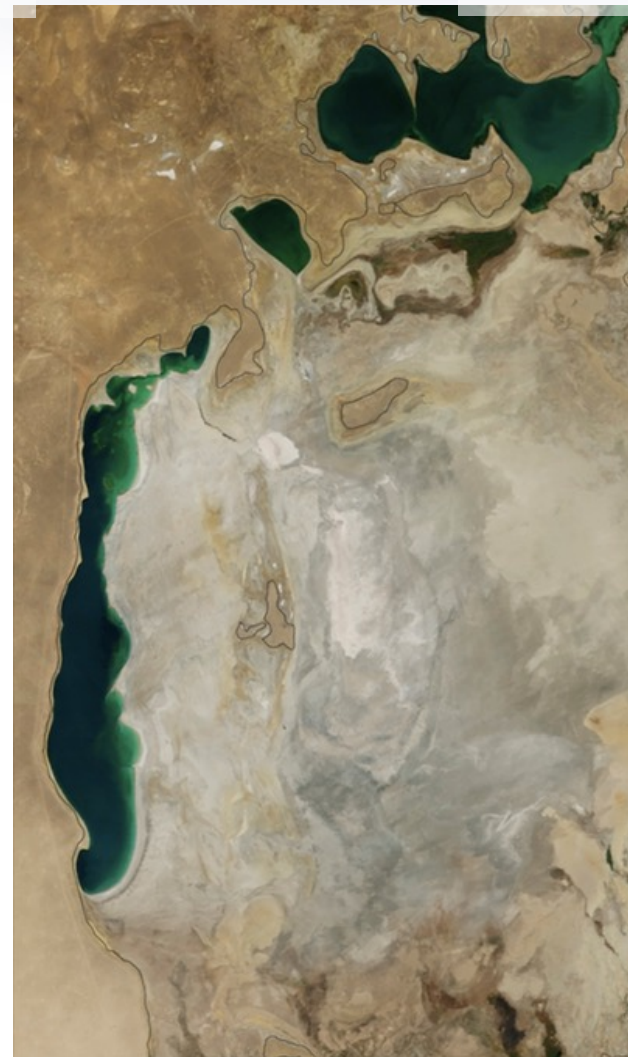
July - September, 1989

2003



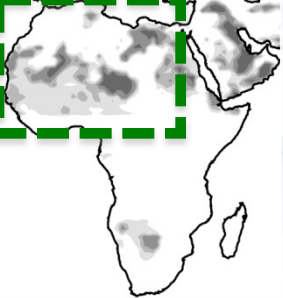
August 12, 2003

2014



Aral Sea





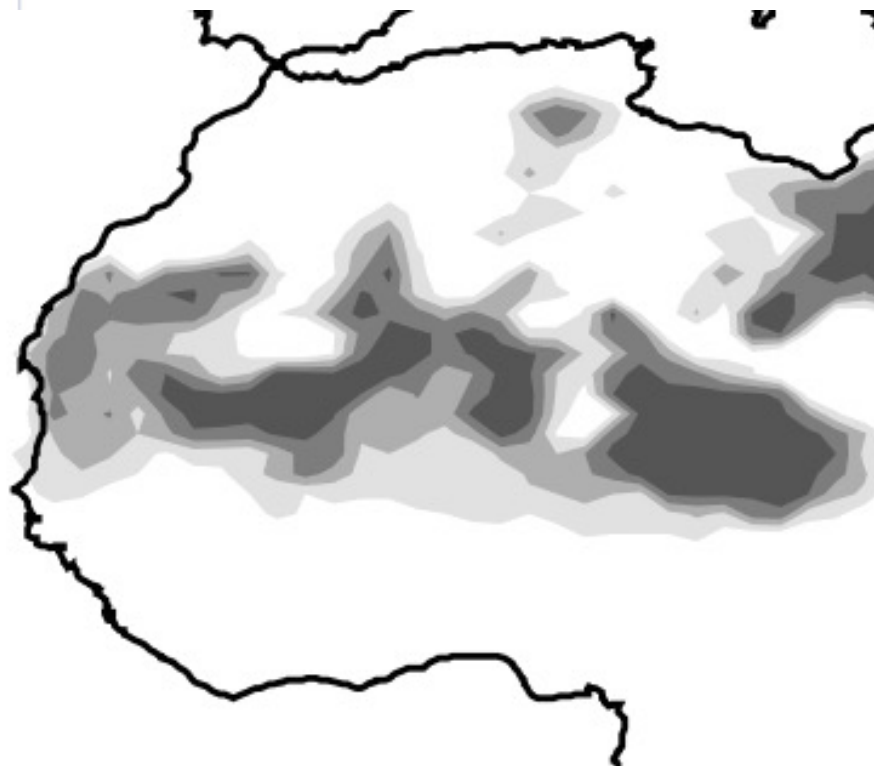
Sahara
Sahel



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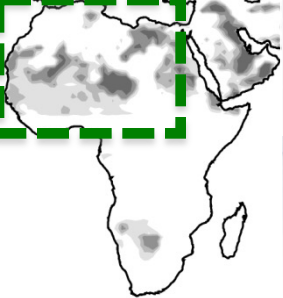
Aemet
Agencia Estatal de Meteorología



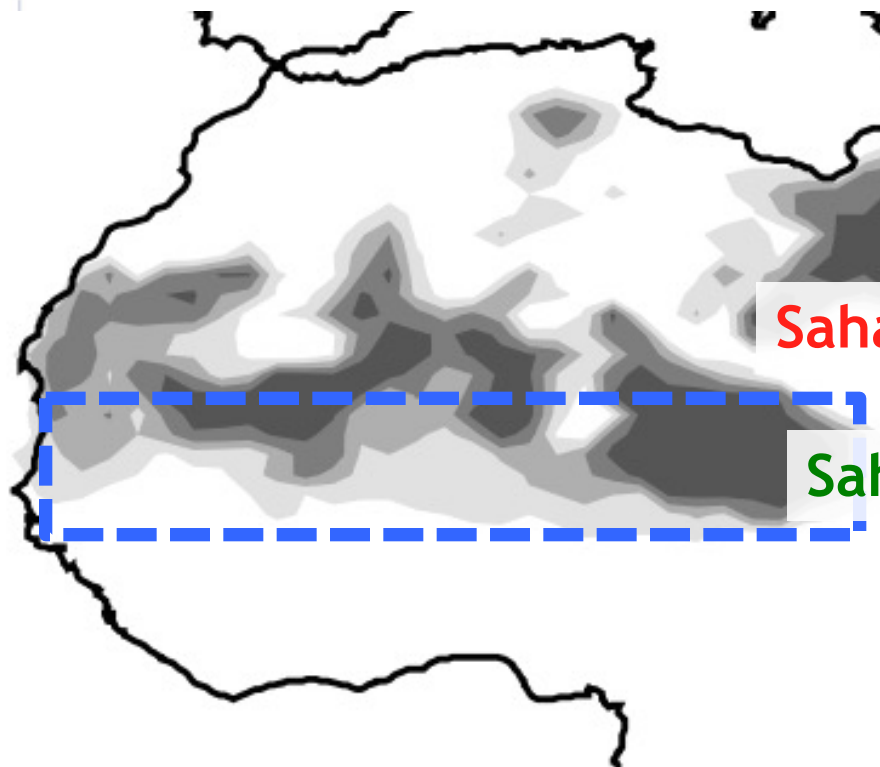
Detección satélite

of Days AI > 0.7 or 1





Sahara
Sahel



Sahara

Sahel



Hyper arid, < 200 mm

Seasonal rainfalls,
moonzon

18°N

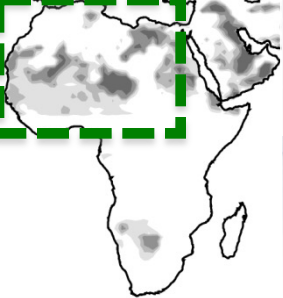
12°N



Detección satélite

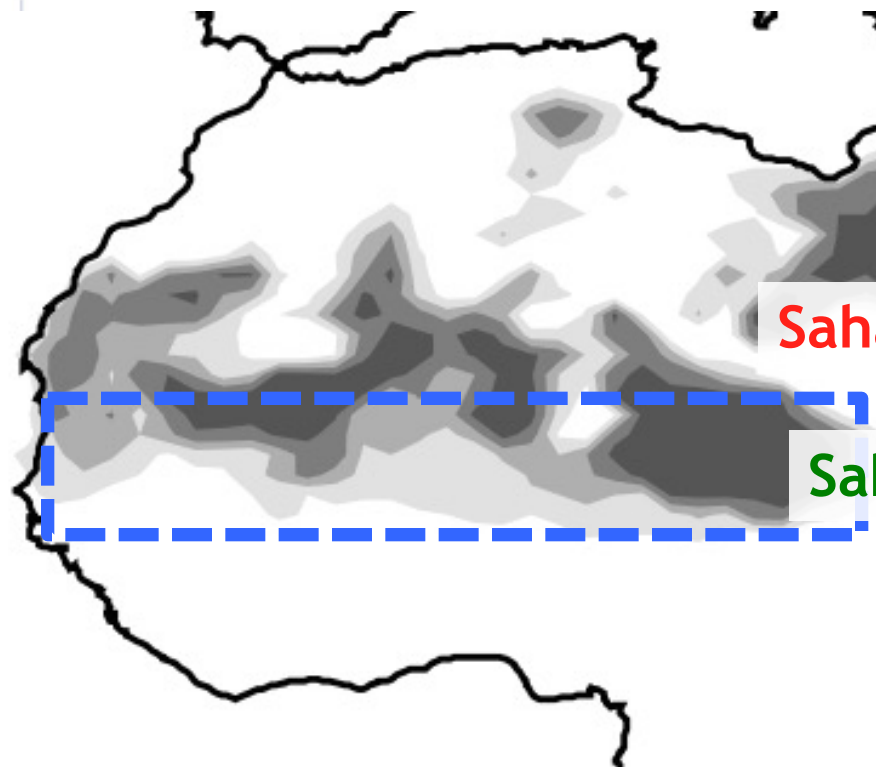
of Days AI > 0.7 or 1





Sahara
Sahel

Sahel

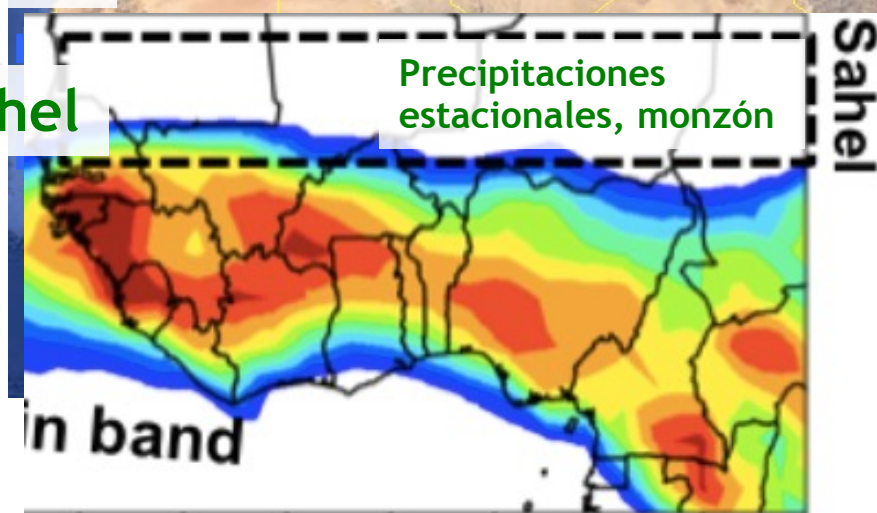


Sahara

Sahel



Hiperárido, < 200 mm



Precipitaciones estacionales, monzón

Sahel

18°N

12°N

in band

10W

0

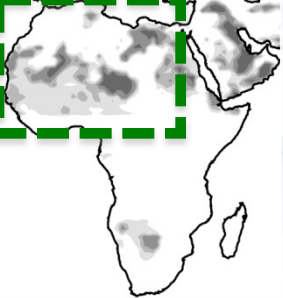
10E

20E


Detección satélite

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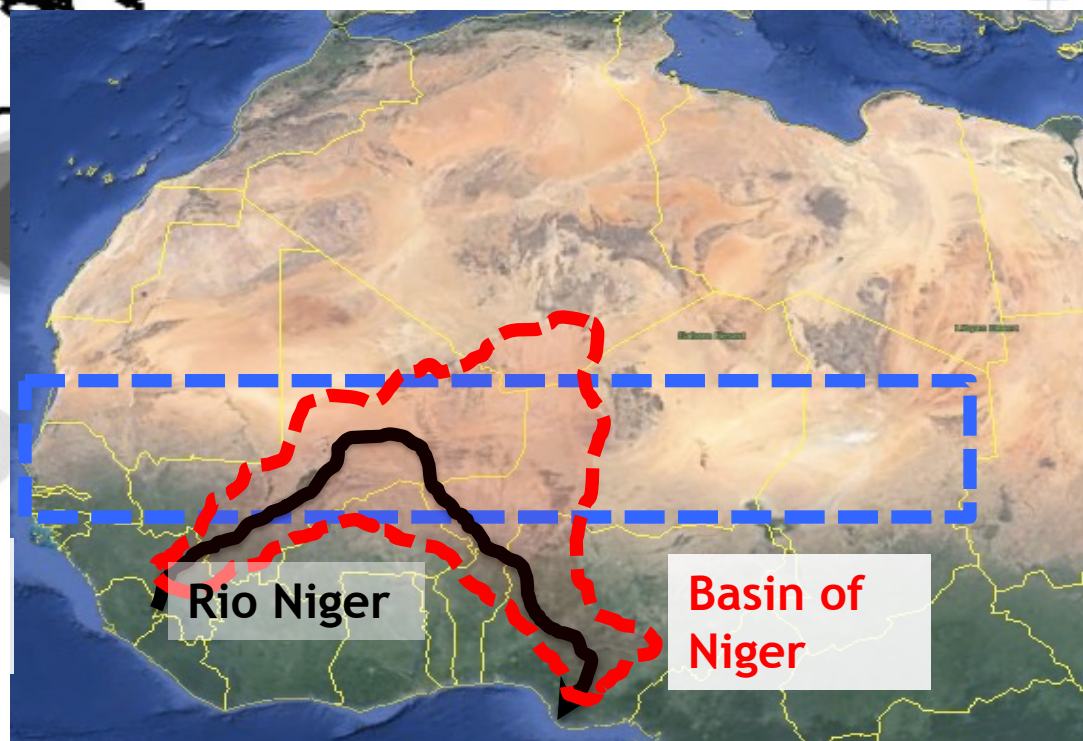
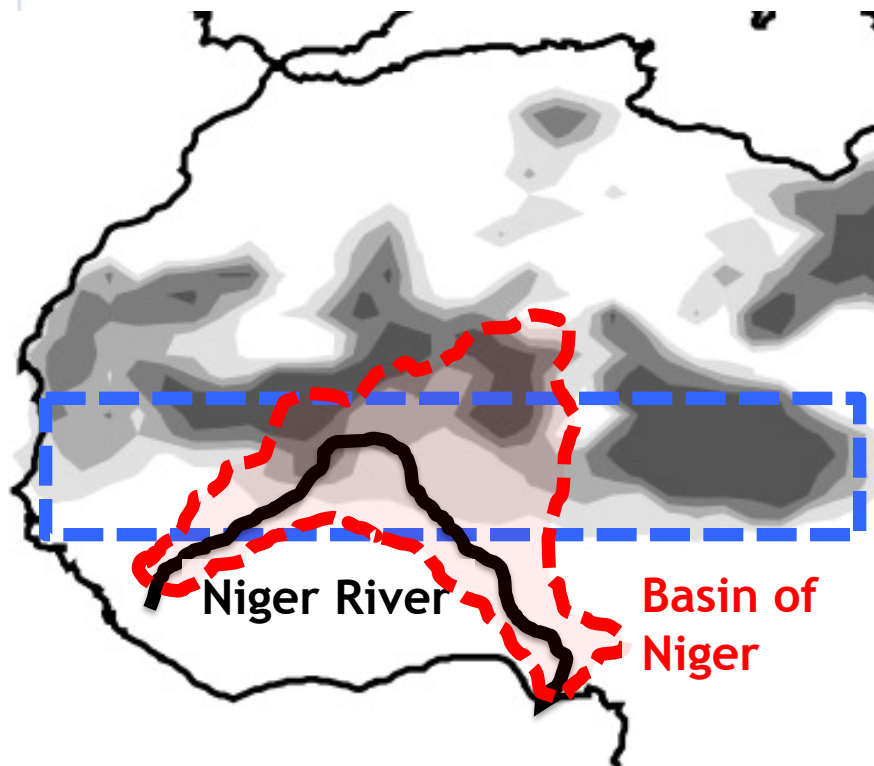
Sahara
Sahel



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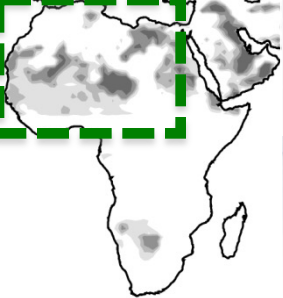
Aemet
Agencia Estatal de Meteorología

Sahel



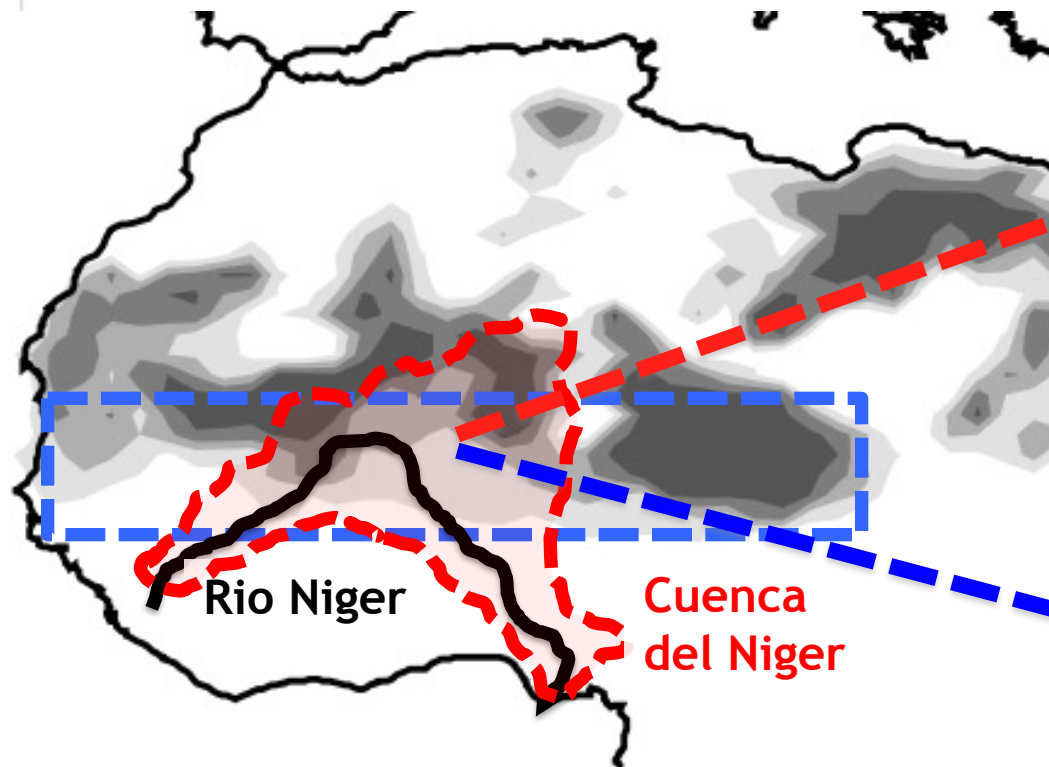
Remote Sensing
satellite





Sahara
Sahel

Sahel



Dry season



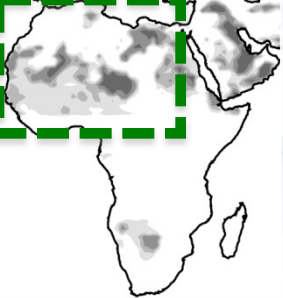
Wet Season



↑
Detección satélite

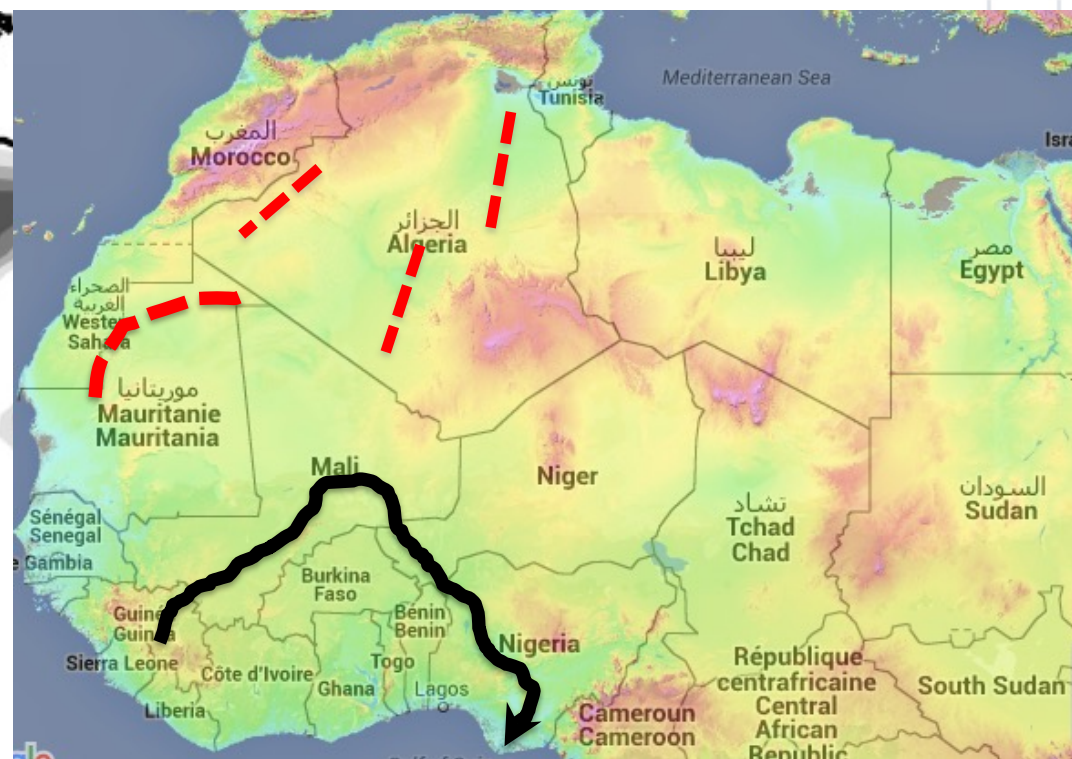
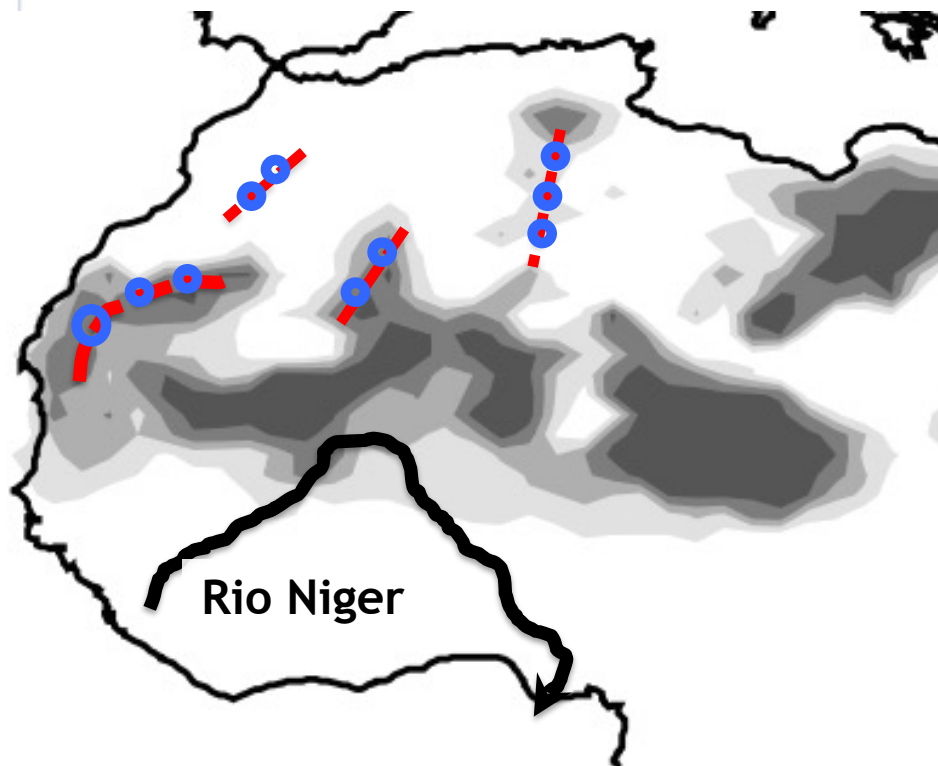
of Days AI > 0.7 or 1





Sahara
Sahel

Sahara



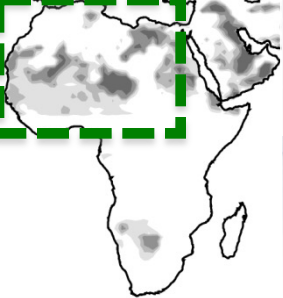
-- bajas topográficas
Wakis: barrancos con inundaciones estacionales

○ chots, sabkas: lechos salados de lagos ecos

Detección satélite

of Days AI > 0.7 or 1





Sahara
Sahel



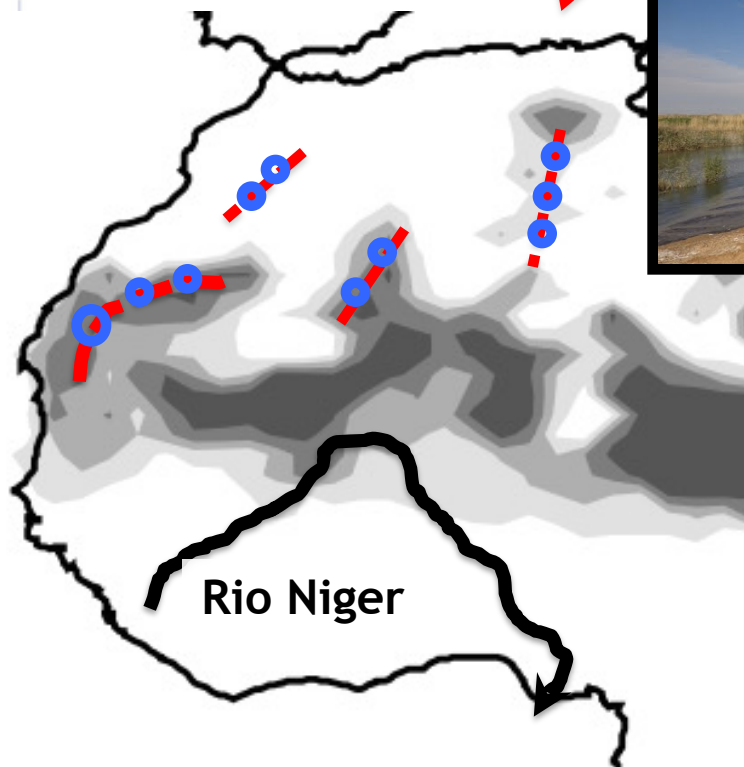
MINISTERIO
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AEMet
Agencia Estatal de Meteorología

Cuenca Ouargla

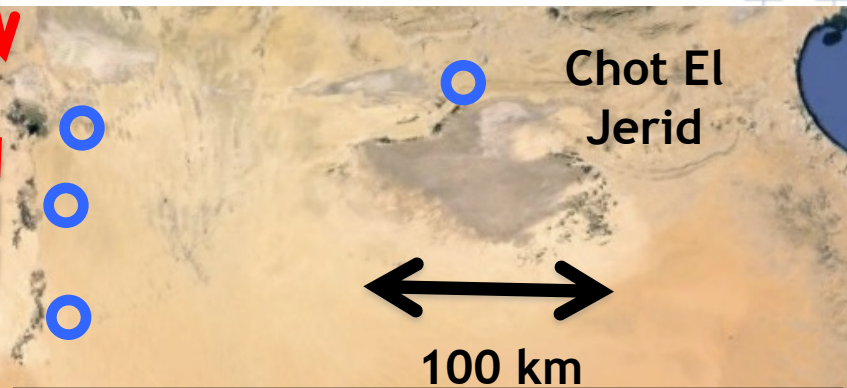
Sahara

chots



-- bajas topográficas
Wakis: barrancos con inundaciones estacionales

○ chotts, sabkhas: lechos salados de lagos ecos



Chot El
Jerid

100 km



dust, aerosols and pollutants

chotts, sabkhas, wadis, salares

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1 and 20 μm
agglomerates

Chotts, Sabkhas



Dry lakes beds



wadis



dust, **aerosols and pollutants**

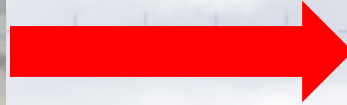
in-situ observations

PM_{10} and $PM_{2.5}$ levels

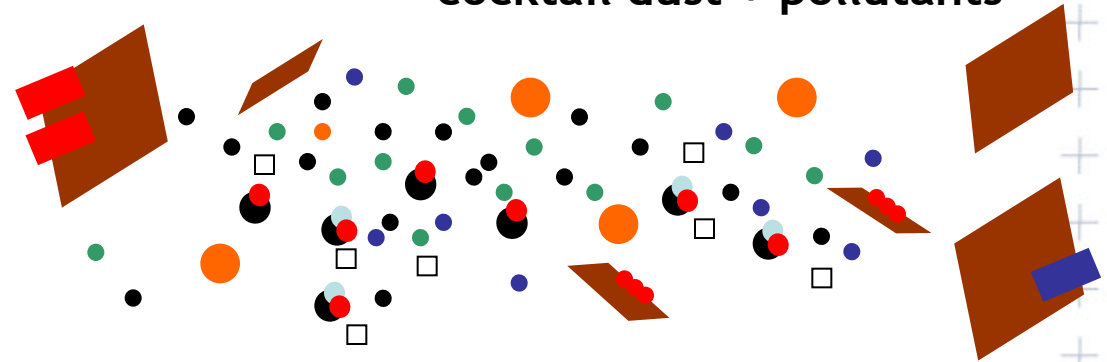
PM_{10} and $PM_{2.5}$ composition

complementary observations

observation network



people live in cities and breath a cocktail dust + pollutants



aerosols, a cocktail of chemicals:

dust

sulphate

nitrate

organic matter

black carbon (soot)

metals (Ni, As, Cd, V, Co...)

sea salt

size: 1 nm (10^{-9} m) to 20 μ m (10^{-6} m)

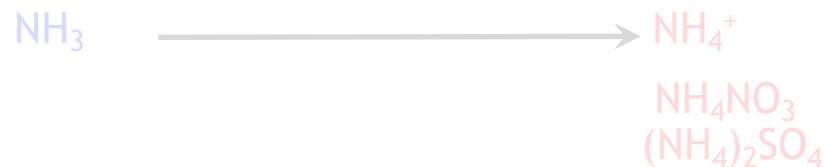
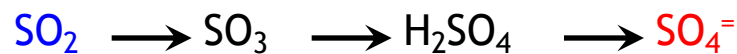
human hair: 70 μ m

aerosols, a cocktail of chemicals:

- dust
- sulphate
- nitrate
- organic mater
- black carbon (soot)
- metals (Ni, As, Cd, V, Co...)
- sea salt

gas precursor

aerosol

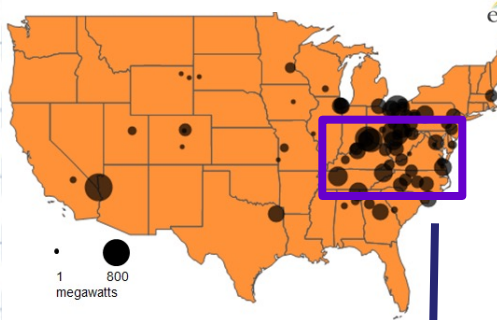


SO_2 : oil refineries, coal power plants, ships, industry

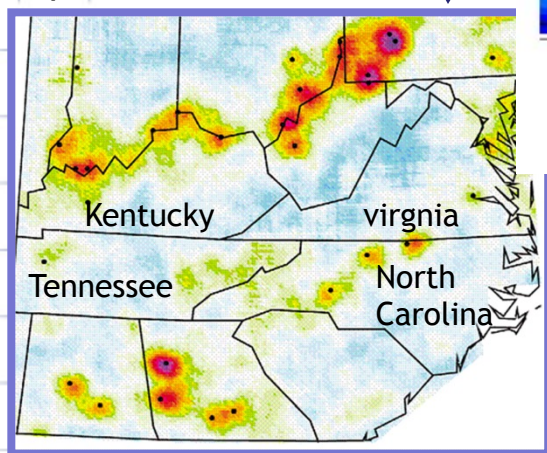
sulfato

122 Tg/y

coal power plants

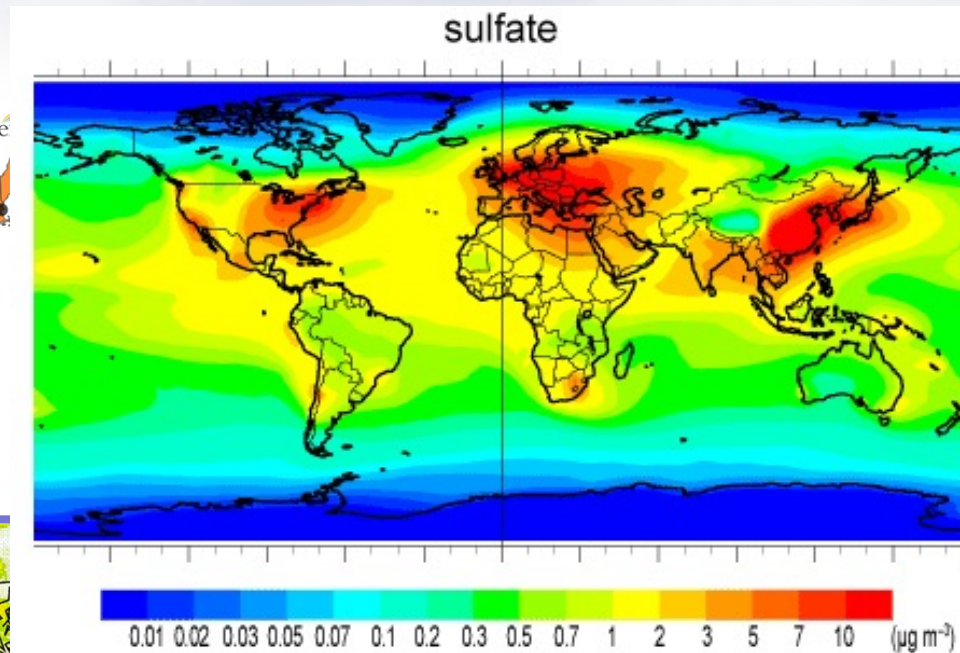


promedio 2005-2007



OMI SO₂ DU

-0.3 0.0 0.3 0.6 DU



coal power plants



VOL. 15, No. 4

JOURNAL OF CLIMATE

15 FEBRUARY 2002

Single-Scattering Albedo and Radiative Forcing of Various Aerosol Species with a Global Three-Dimensional Model

TOSHIHIKO TAKEMURA* AND TERUYUKI NAKAJIMA
OLEG DUBOVIK, BRENT N. HOLBEN, AND STEFAN KINNE

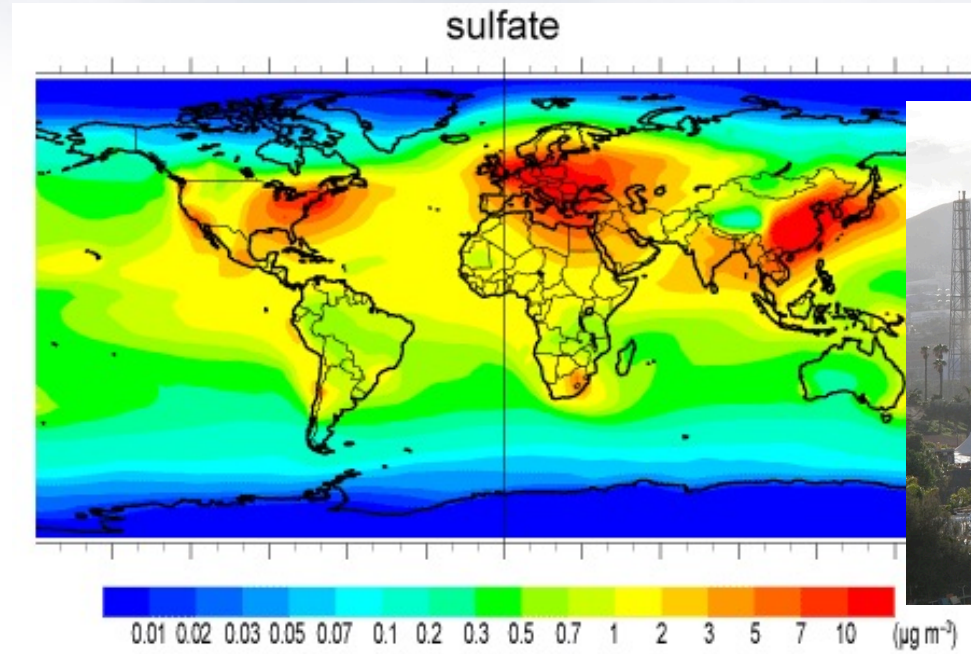
GEOPHYSICAL RESEARCH LETTERS, VOL. 38, L21811, doi:10.1029/2011GL049402, 2011

Estimation of SO₂ emissions using OMI retrievals

V. E. Fioletov, C. A. McLinden, N. Krotkov, M. D. Moran, and K. Yang

sulfato

Oil refinery



Oil refinery



ships

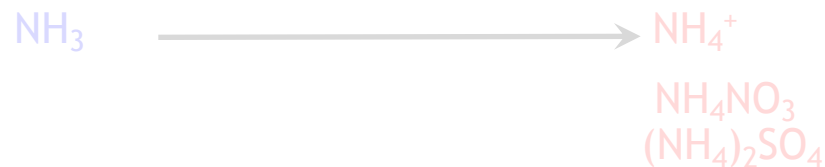


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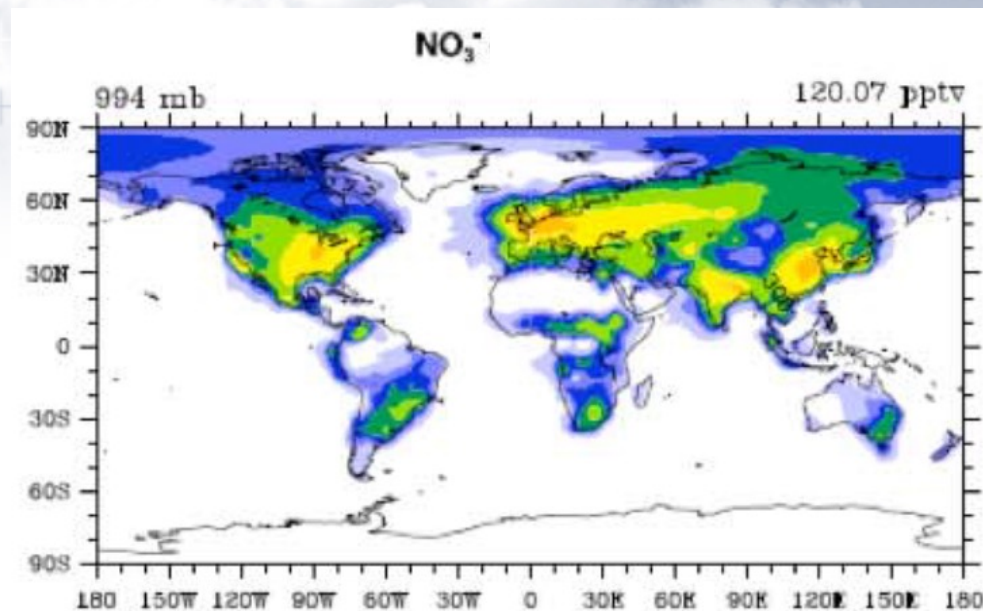


NO_x : vehicle exhaust, power plants, industry

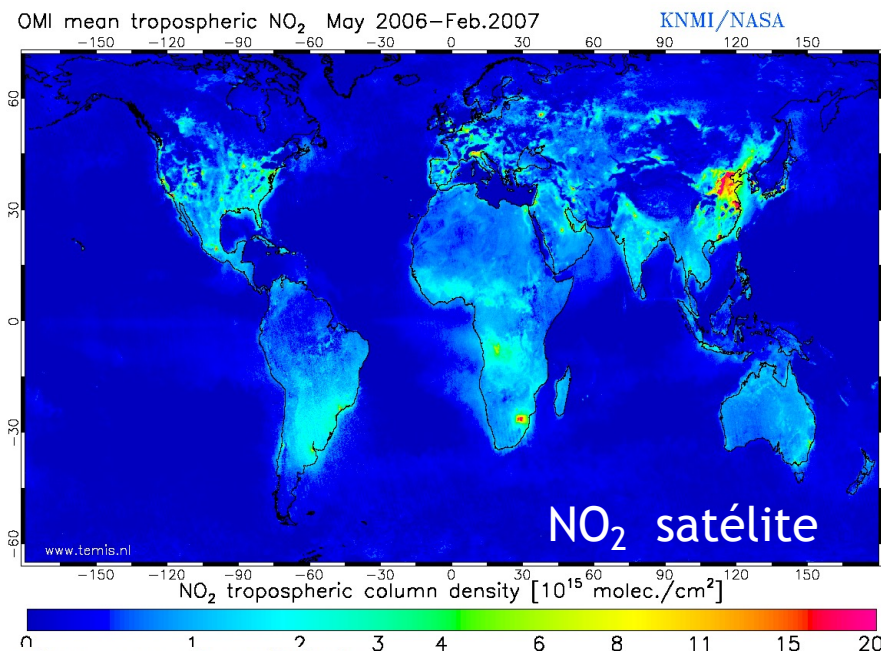


nitrate

18 Tg/y



NH_4NO_3

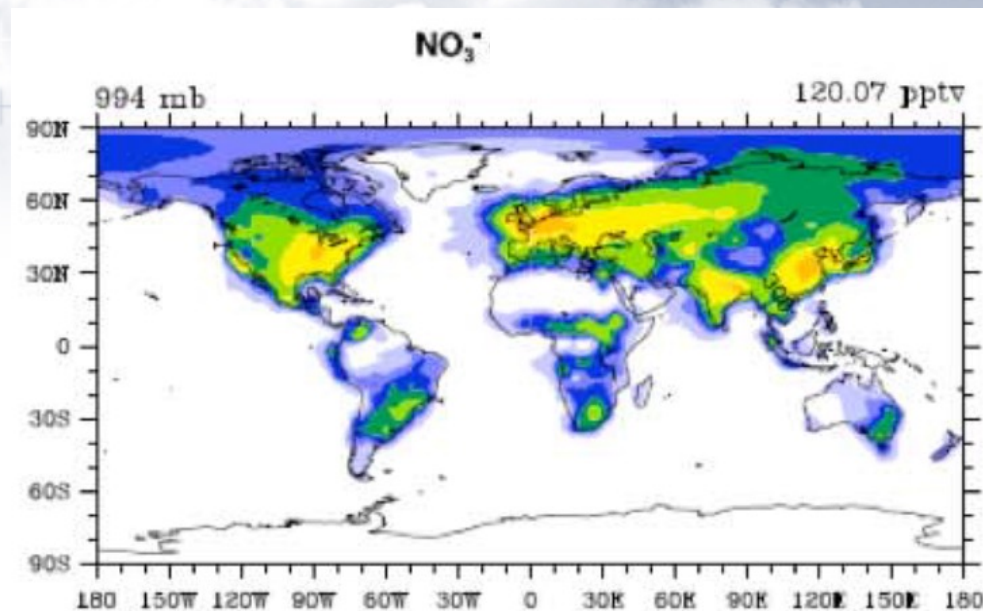


Atmos. Chem. Phys., 12, 9479–9504, 2012
www.atmos-chem-phys.net/12/9479/2012/

L. Xu and J. E. Penner

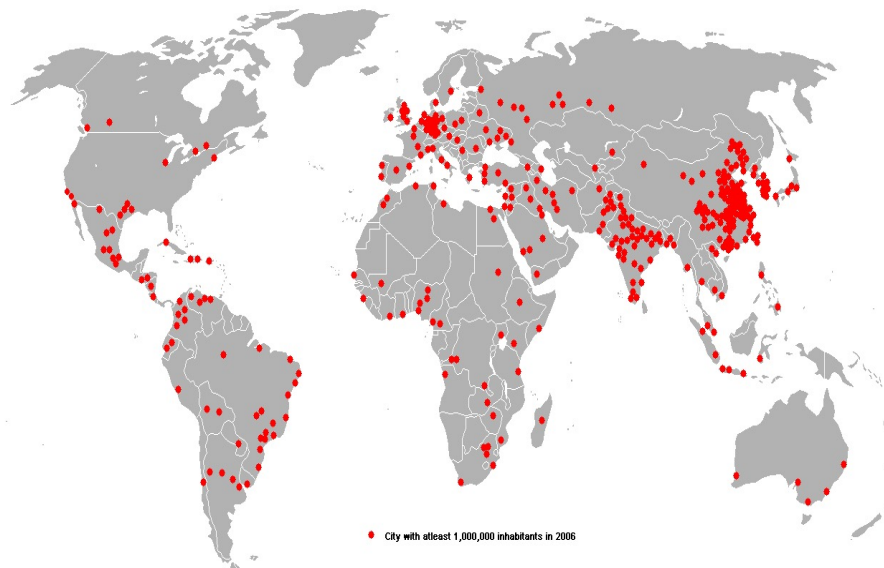
Global simulations of nitrate and ammonium aerosols and their radiative effects

nitrate



NH_4NO_3

OMI mean tropospheric NO_2 May 2006–Feb. 2007 KNMI/NASA



cities > 1 million inhabitants

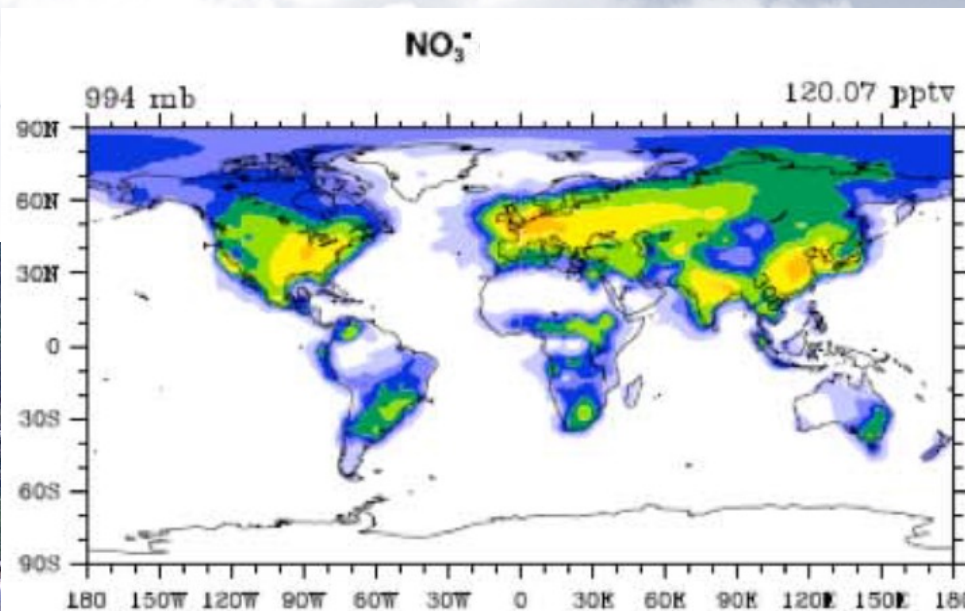
Atmos. Chem. Phys., 12, 9479–9504, 2012
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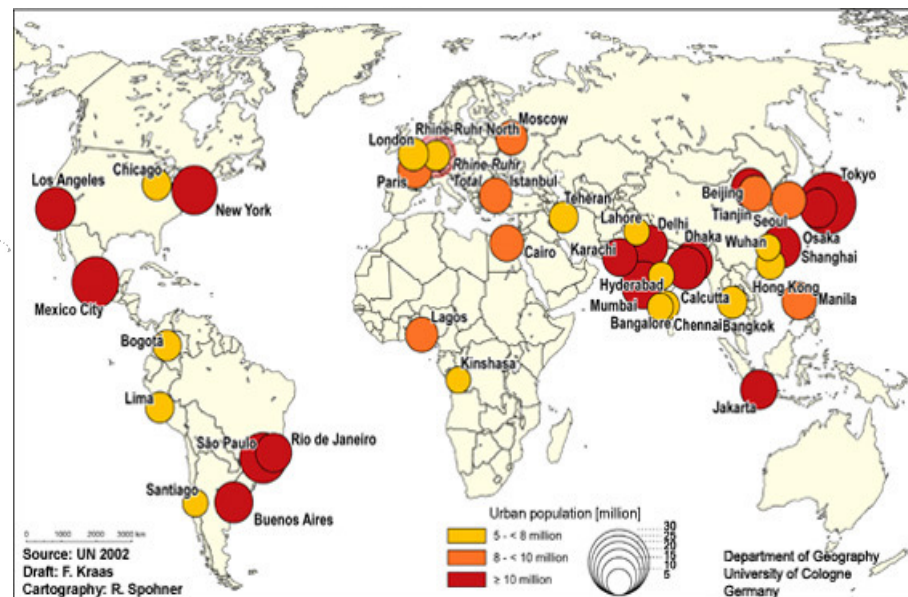
nitrato

NH_4NO_3



OMI mean tropospheric NO_2 May 2006–Feb.2007

KNMI/NASA



Mega-ciudades, > 5
Millones habitantes

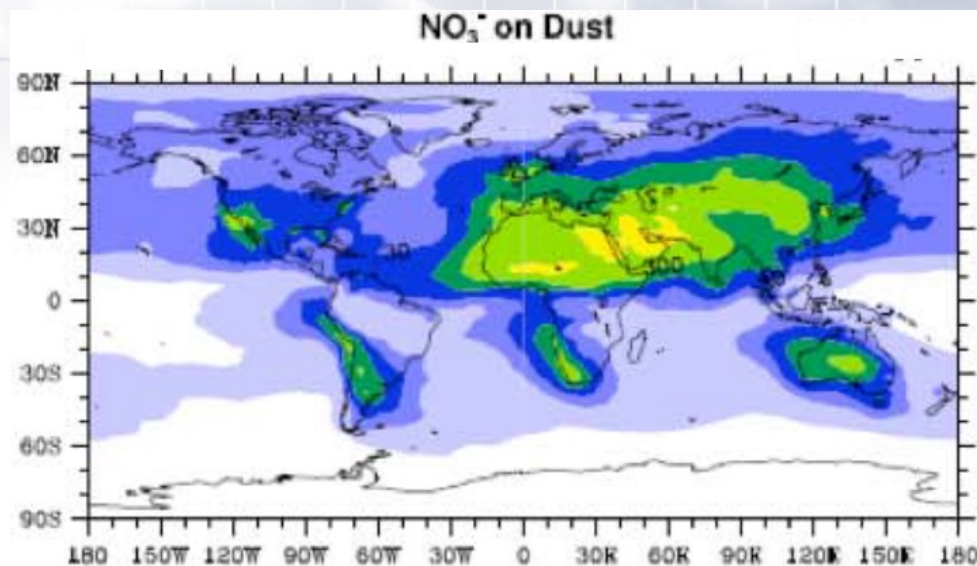


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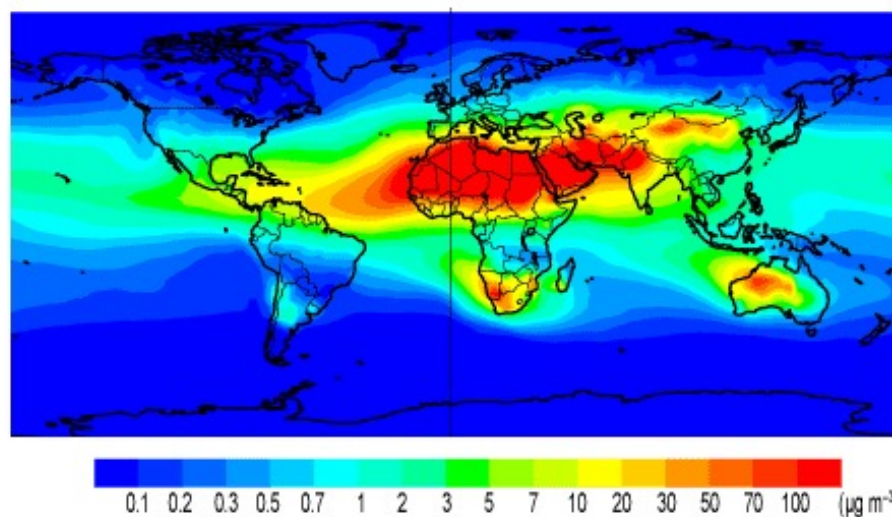
Global simulations of nitrate and ammonium aerosols and their
radiative effects

nitrate



NO_3^- - dust
 $\text{Ca}(\text{NO}_3)_2$

soil dust



Atmos. Chem. Phys., 12, 9479–9504, 2012
www.atmos-chem-phys.net/12/9479/2012/

L. Xu and J. E. Penner

Global simulations of nitrate and ammonium aerosol
radiative effects

<http://www.knmi.nl/omi/research/product/index.php>

aerosols, a cocktail of chemicals:

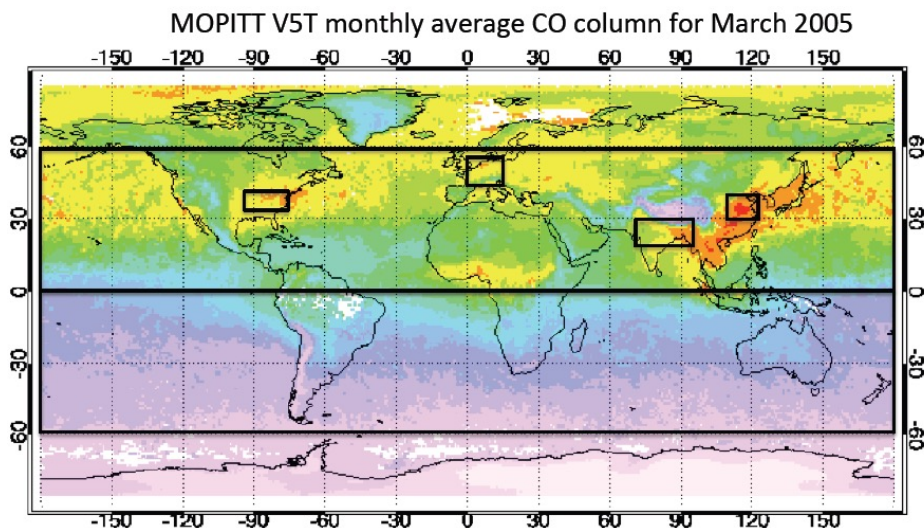
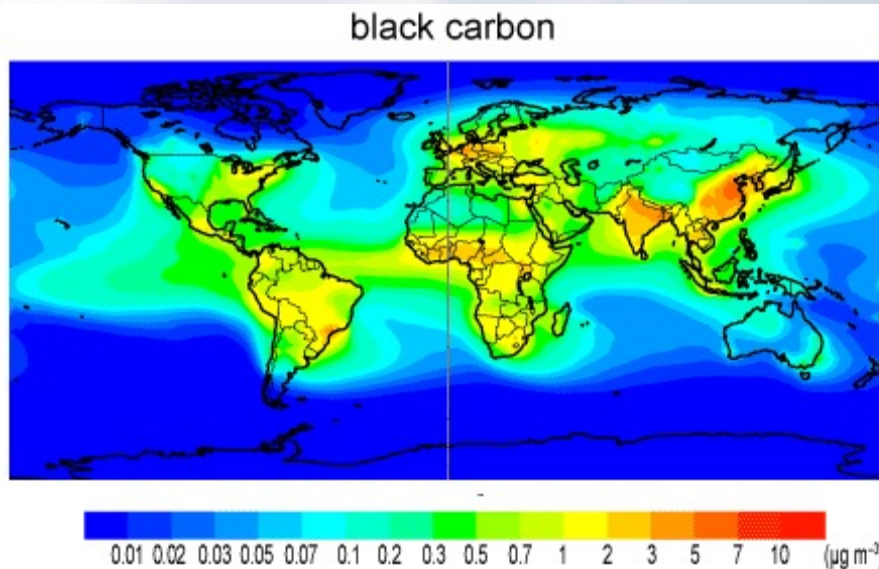
{
dust
sulphate
nitrate
organic mater
black carbon (soot)
metals (Ni, As, Cd, V, Co...)
sea salt

Black carbon: vehicle exhaust (diesel) , combustion sources

black carbon

10.5 Tg/y

diesel, 4x4, camiones



automóviles



India

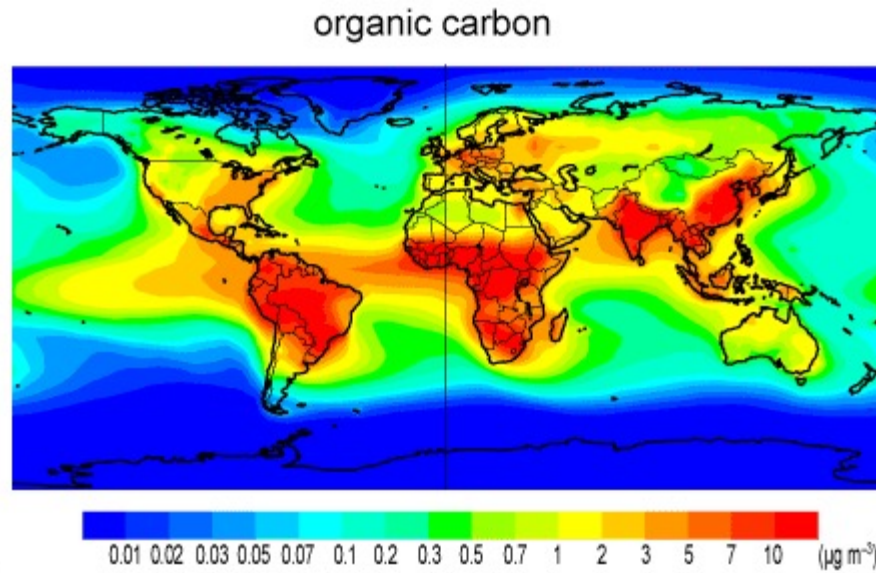


aerosols, a cocktail of chemicals:

{
dust
sulphate
nitrate
organic mater
black carbon (soot)
metals (Ni, As, Cd, V, Co...)
sea salt

organic matter: combustion sources, vehicle exhaust

carbano orgánico

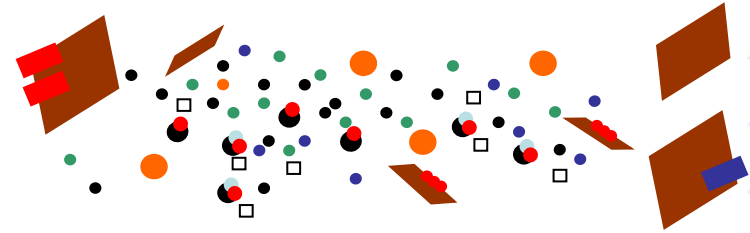


automóviles





people live in cities and breath a cocktail dust + pollutants

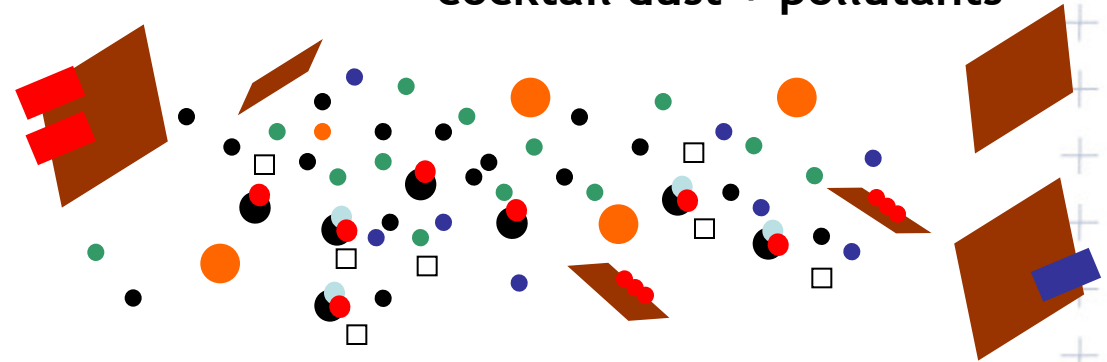


© Mac Mackay photo





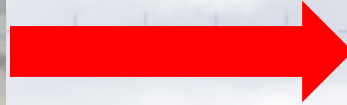
people live in cities and breath a cocktail dust + pollutants



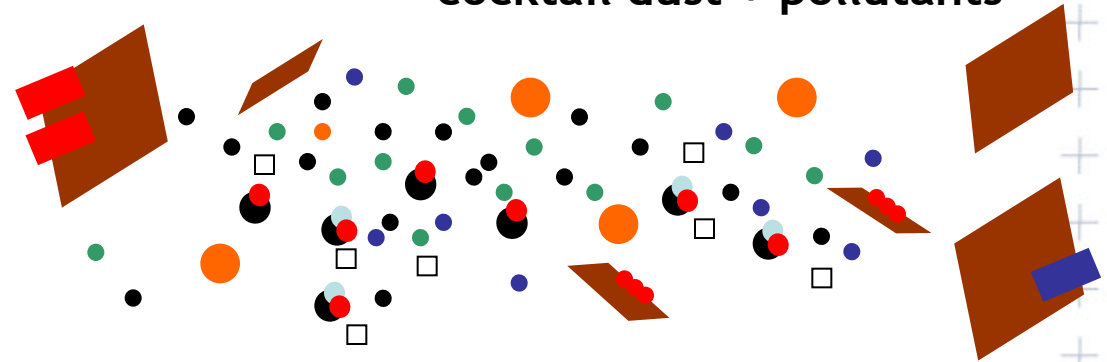
In air quality, aerosols:

PM₁₀: mass concentration ($\mu\text{g}/\text{m}^3$) of all aerosols smaller than 10 μm
inhalable particles

PM_{2.5}: mass concentration ($\mu\text{g}/\text{m}^3$) of all aerosols smaller than 2.5 μm
alveolar particles



people live in cities and breath a cocktail dust + pollutants



PM_{10} : dust + sea salt + (sulphate + nitrate + organic matter + black carbon + metals) ..

$PM_{2.5}$: dust + sea salt + (sulphate + nitrate + organic matter + black carbon + metals) ..

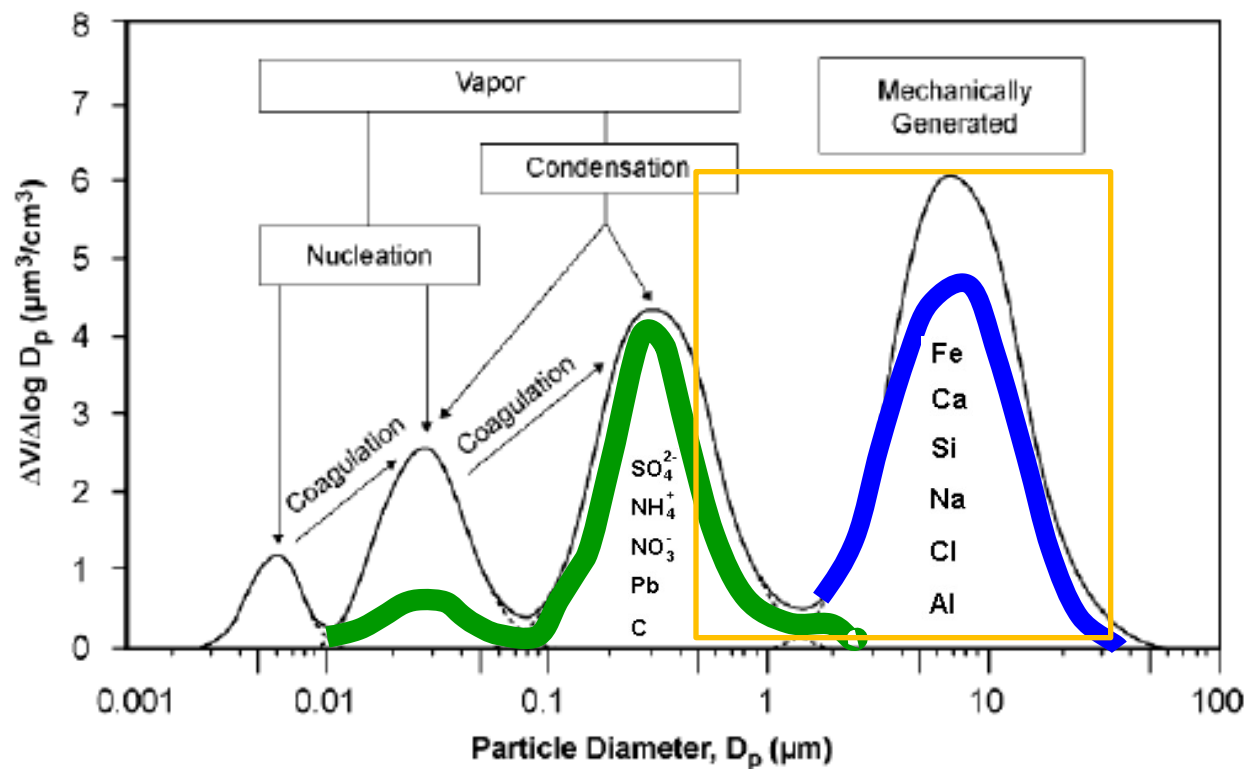
PM_{10} : dust + sea salt + vehicle exhaust + oil refining + power plants + ships +...

$PM_{2.5}$: dust + sea salt + vehicle exhaust + oil refining + power plants + ships +...

PM₁₀ (diameter <10 microm)

PM_{2.5}

PM_{2.5-10}



ultrafine
<0.1 μm

accumulation
0.1 - 1 μm

Coarse
1 - 10 μm

Mineral dust :

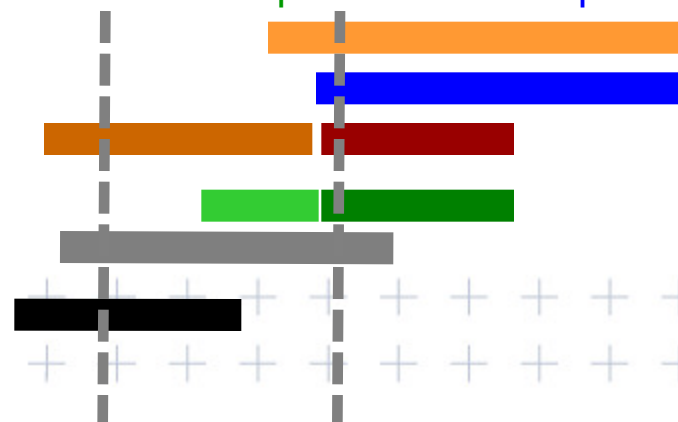
Marine salt:

Sulfate:

Nitrate:

Organic aerosol:

black carbon:



dust, aerosols and pollutants

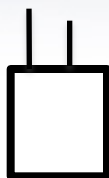
in-situ observations

PM_{10} and $PM_{2.5}$ levels

PM_{10} and $PM_{2.5}$ composition

complementary observations

observation network



dust - air quality



What to measure to protect human health ?

dust, aerosols and pollutants

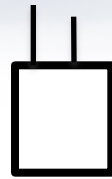
in-situ observations

PM₁₀ and PM_{2.5} levels

PM₁₀ and PM_{2.5} composition

complementary observations

observation network



dust air quality



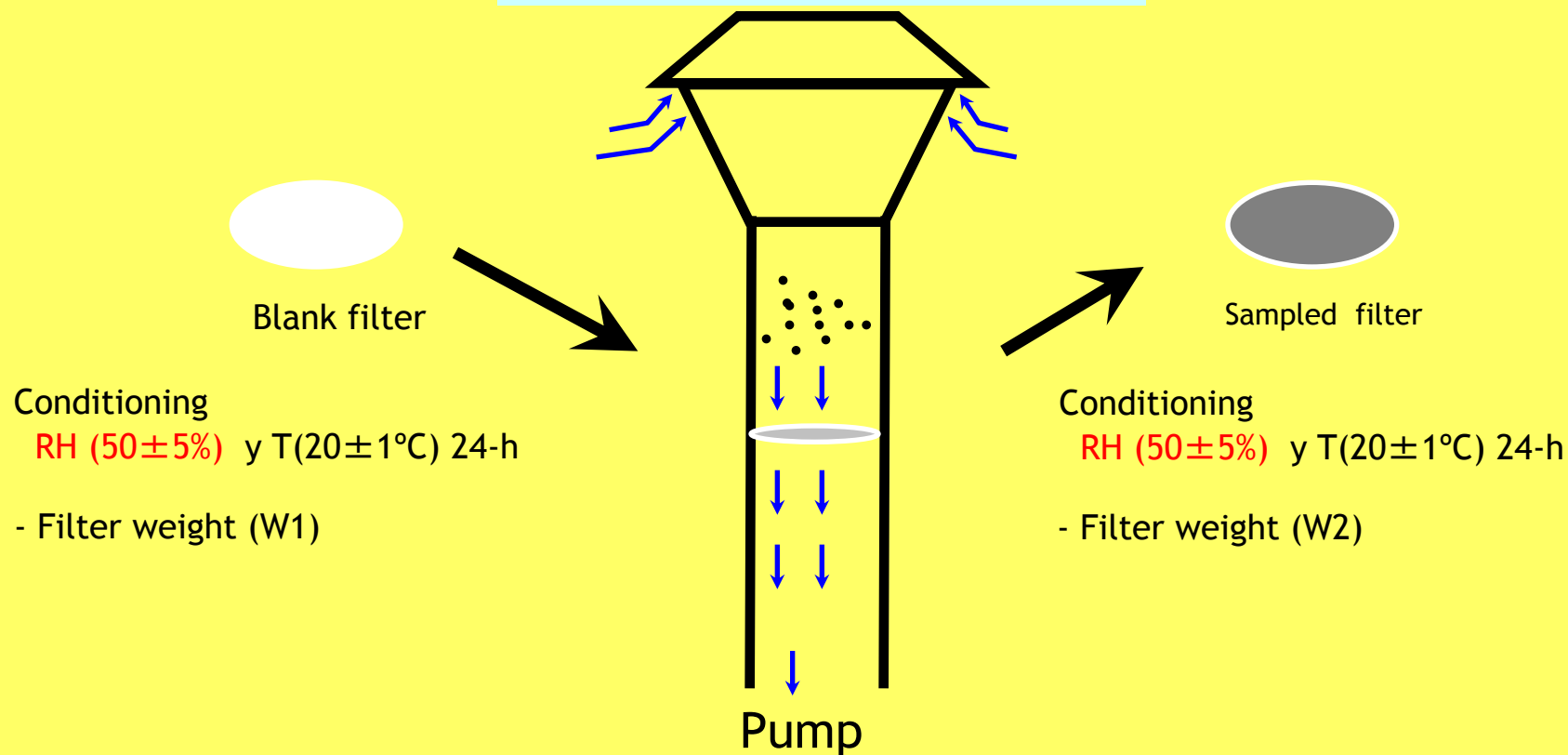
1. PM_{10} and $PM_{2.5}$ levels

-method-01 manual gravimetry method



-method-01: reference - manual gravimetry

$$PM = \frac{(W2 - W1)}{\text{Volume}} \mu\text{g}/\text{m}^3$$



It is recommended to use standardised protocols
national standard method
or already existing international standard methods

- PM₁₀ and PM_{2.5} sampler**
- sampling procedure**
- weighing procedure**

example:

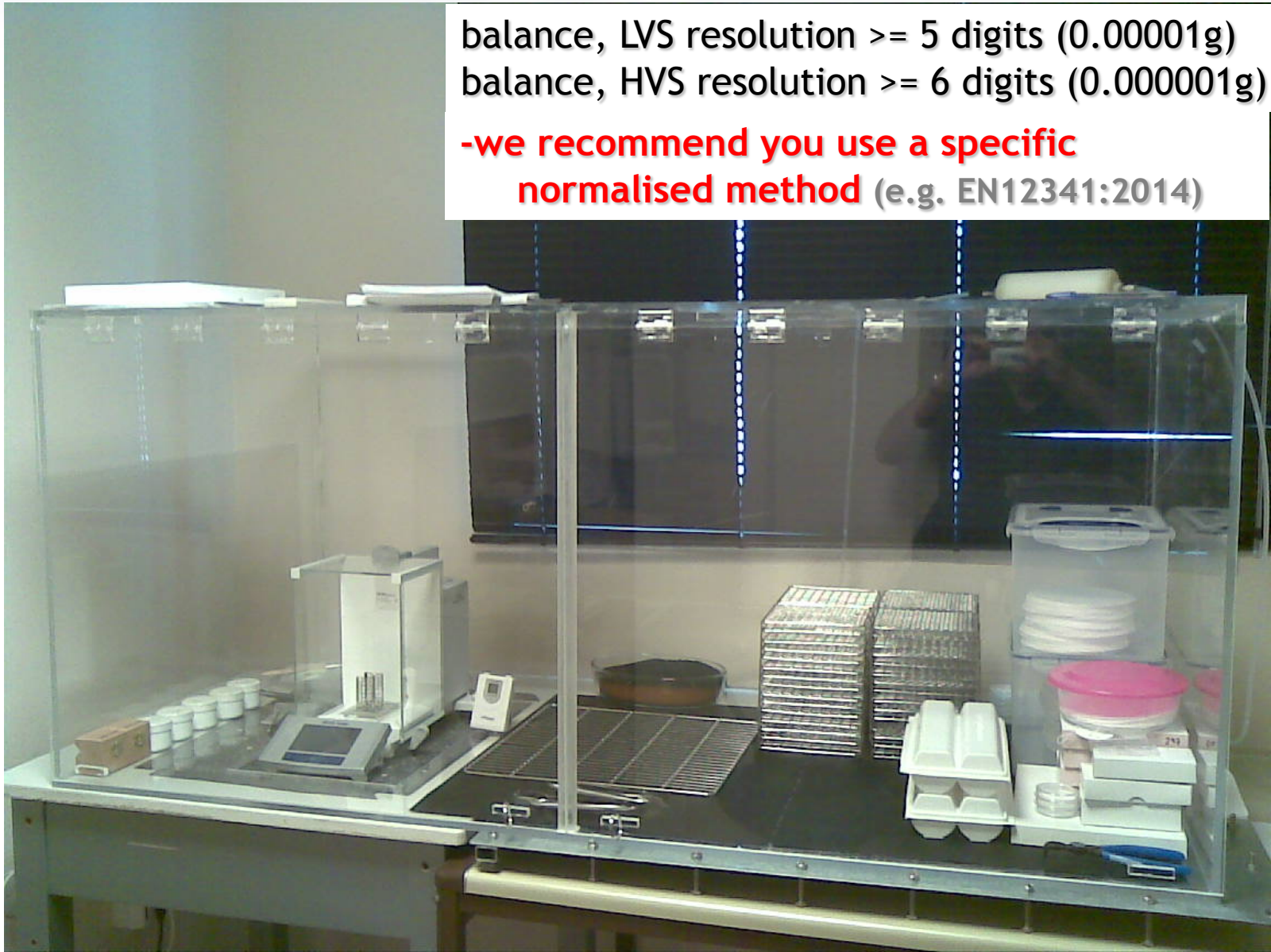
EN 12341:2014

**Ambient air. Standard gravimetric measurement method for the
determination of the PM₁₀ or PM_{2.5} mass concentration of suspended
particulate matter**

Room for weighting the filters: RH =50% (30 %) and 20°C

balance, LVS resolution ≥ 5 digits (0.00001g)
balance, HVS resolution ≥ 6 digits (0.000001g)

**-we recommend you use a specific
normalised method (e.g. EN12341:2014)**



PM_{10}
Blank filter

PM_{10}
sample urban air

PM_{10}
sample in dust days



-we recommend you use a specific
normalised method (e.g. EN12341:2014)

Filters: Quartz, Teflon, Cellulose

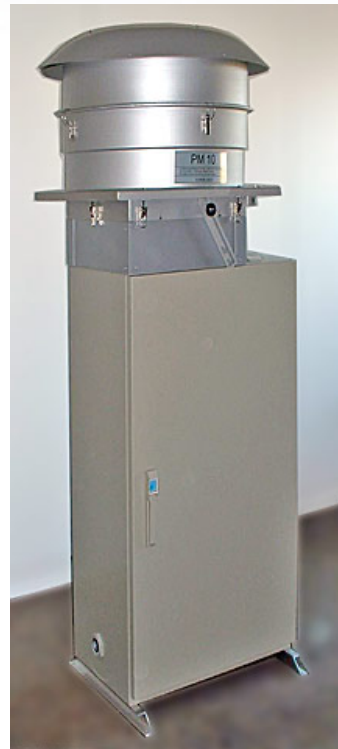
Low Volume Sampler

LVS: **2.3 m³/h**



High Volume Sampler

HVS: **68 m³/h**



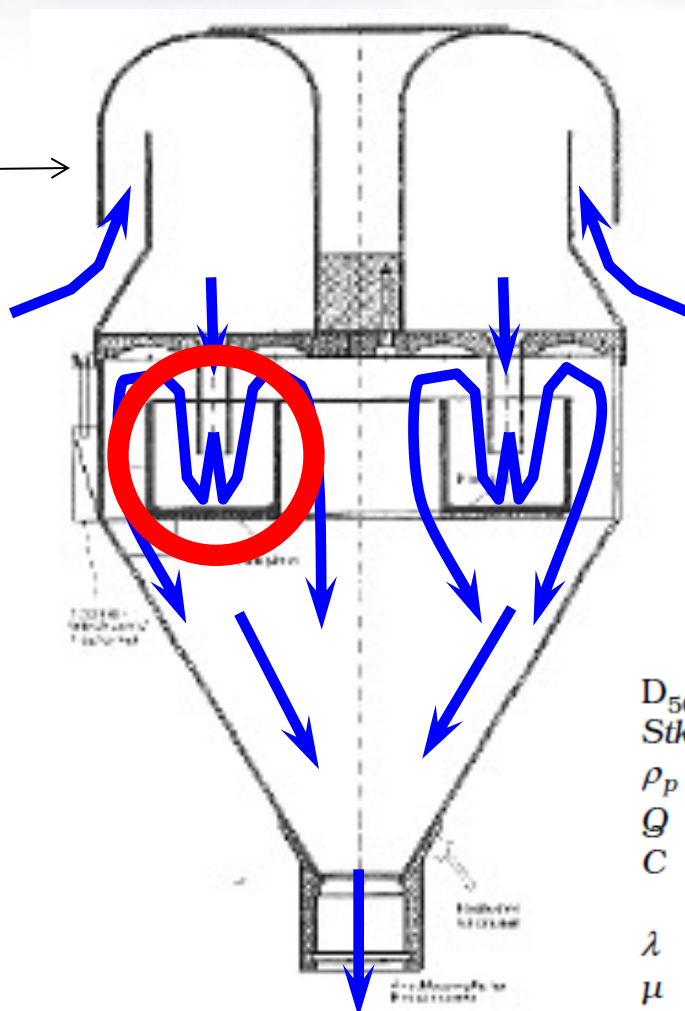
HVS: **30 m³/h**



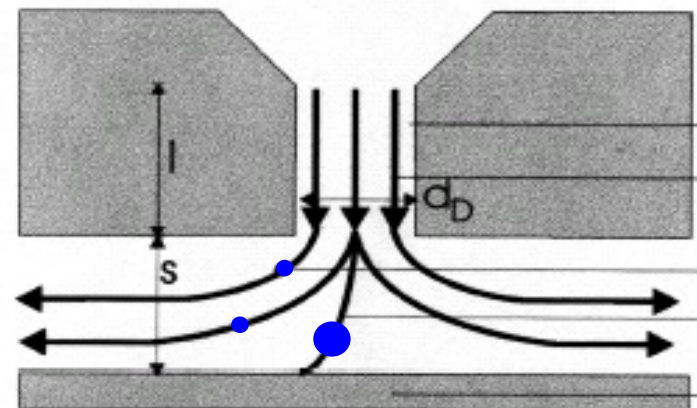
-we recommend you use a specific normalised method (e.g. EN12341:2014). Ask to the distributor if the sampler is designed to any standards

Inlets, airflows....

PM₁₀, PM_{2.5}



Filter



$$D_{50} = \sqrt{\frac{9\pi Stk \mu W^3}{4\rho_p CQ}}$$

D_{50} = particle cut-point diameter centimeter

Stk = Stokes number = 0.23

ρ_p = particle density (g/cm³)

Q = volumetric flow rate (cm³/s)

C = Cunningham slip correction

$= 1 + 2.492 \lambda/D_{50} + 0.84 \lambda/D_{50} \exp(-0.435 D_{50}/\lambda)$

λ = gas mean free path

μ = gas viscosity (dyne•s/cm²)

W = nozzle diameter (cm)

The Stokes number is a dimensionless parameter that characterizes impaction.



dust air quality

1. PM_{10} and $PM_{2.5}$ levels

-method-01: reference - manual gravimetry

Manual gravimetry

advantage: reference method

disadvantage: poor time resolution, 24-h average
manual work
takes 3 days to know PM_{10} concentration



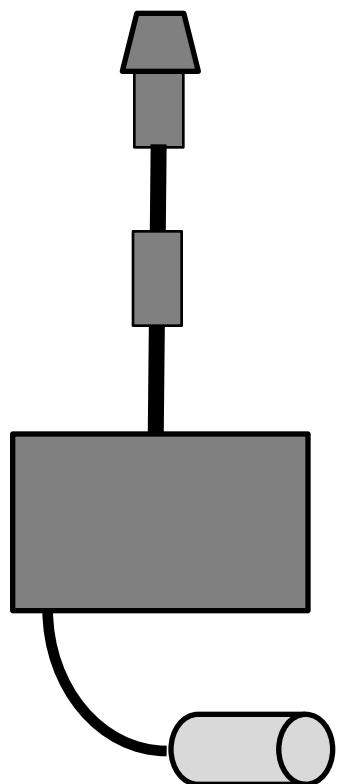
dust air quality



1. PM_{10} and $PM_{2.5}$ levels

- method-01: reference - manual gravimetry
- method-02: automatic

-method-02: automatic



1. Impactor PM_{10} / $PM_{2.5}$

2. RH reductor / heater

3. Sensor

Beta radiation attenuation

4. Pump / Flow meter

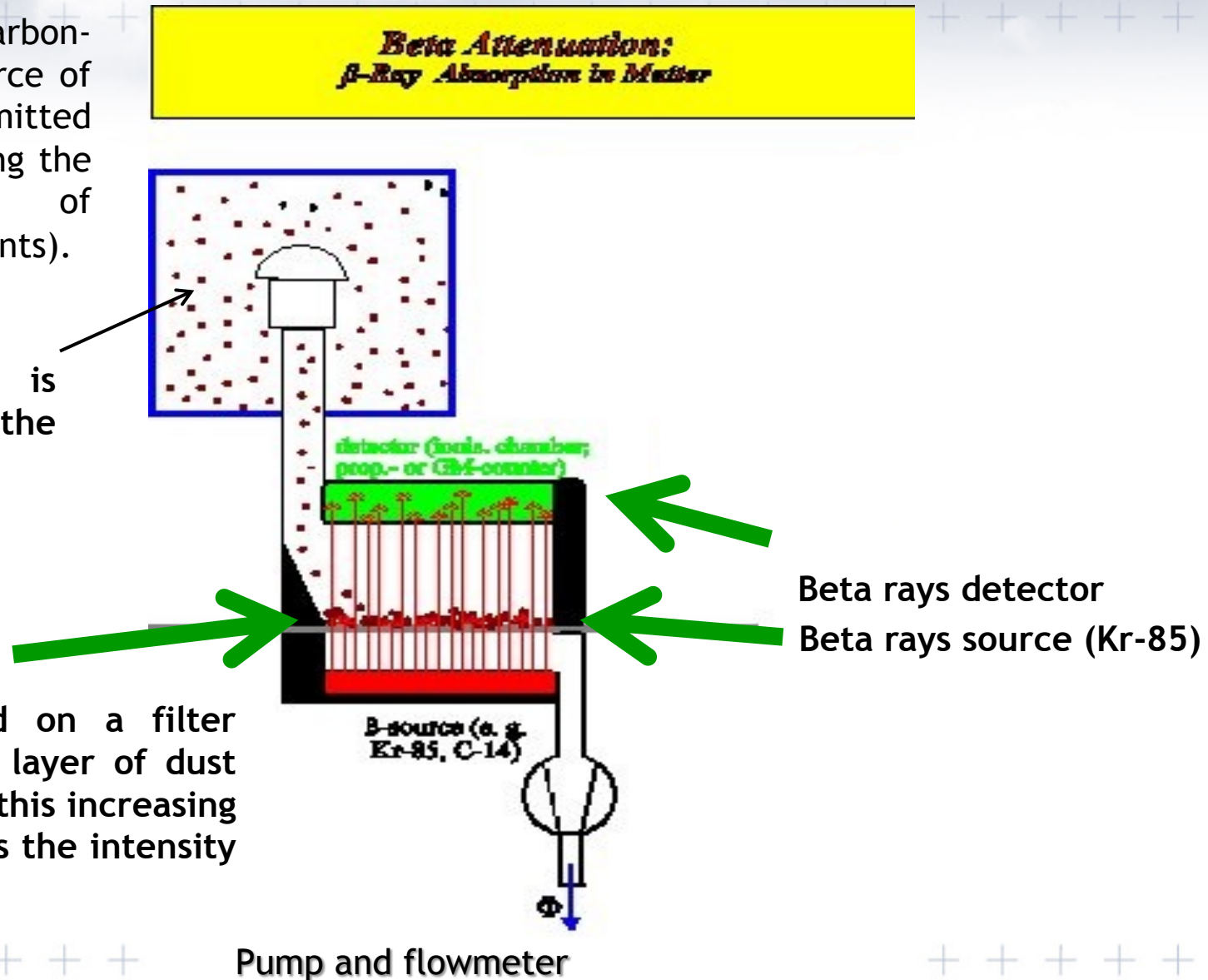
Continuous measurements of PM (PM_{10} , $PM_{2.5}$, PM_1 or TSP)

PM with Beta attenuation

Krypton-85 or Carbon-14 is used as source of beta radiation (emitted by electrons during the nuclear decay of radioactive elements).

Ambient air is drawn through the sample system

Dust is deposited on a filter continuously. The layer of dust is building up and this increasing dust mass weakens the intensity of the beta beam.

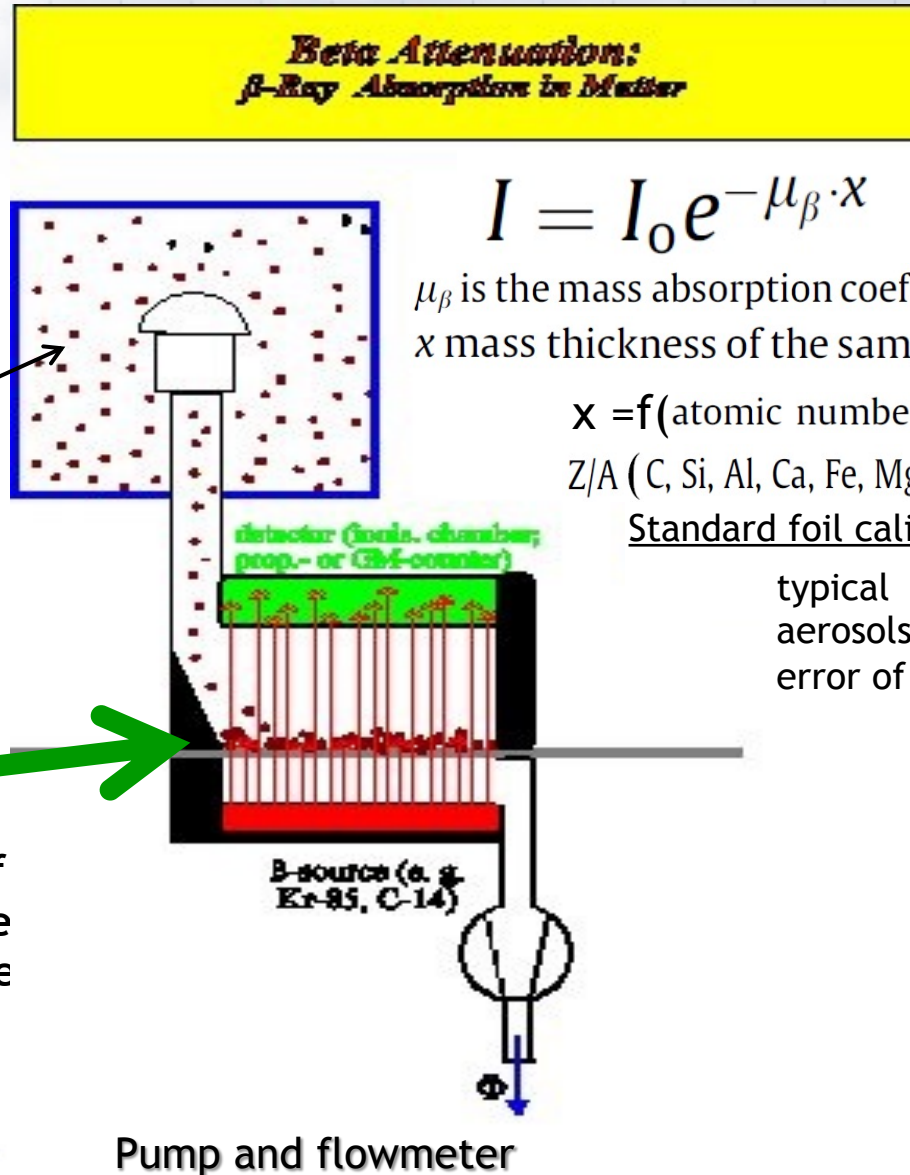


PM with Beta attenuation

Krypton-85 or Carbon-14 is used as source of beta radiation (emitted by electrons during the nuclear decay of radioactive elements).

Ambient air is drawn through the sample system

Dust is deposited on a continuously. The layer of is building up and this incre dust mass weakens the inte of the beta beam.



PM with Beta attenuation (2)

$$m = F_{cal} \ln \left(\frac{I_0}{I} \right)$$

- **m**: increasing particle mass [μg]
- **F_{cal}**: calibration factor
- **I₀** beta ray intensity at empty filter
- **I** beta ray intensity at loaded filter

The intensities I_0 and I are measured with the detector system. F_{cal} has to be measured directly during the calibration procedure. This is accomplished by replacing the filter with the element having a known mass (mass calibration kit)

The mass concentration is calculated from:

$$\text{PM}_{10} \ \& \ \text{PM}_{2.5} \approx c = \frac{m}{Ft}$$

Where:

c: concentration [$\mu\text{g}/\text{m}^3$]

F: measured air flow [m^3/h]

t: time [h]

-method-02: automatic

The most extended method and the most robust for dusty regions

beta



Tapered Oscillating Microbalance

TEOM

Manual change of the filter

there are other methods, but are less robust for dusty regions



Optical Particle Counters

cleaning of optics
laser maintenance



dust air quality

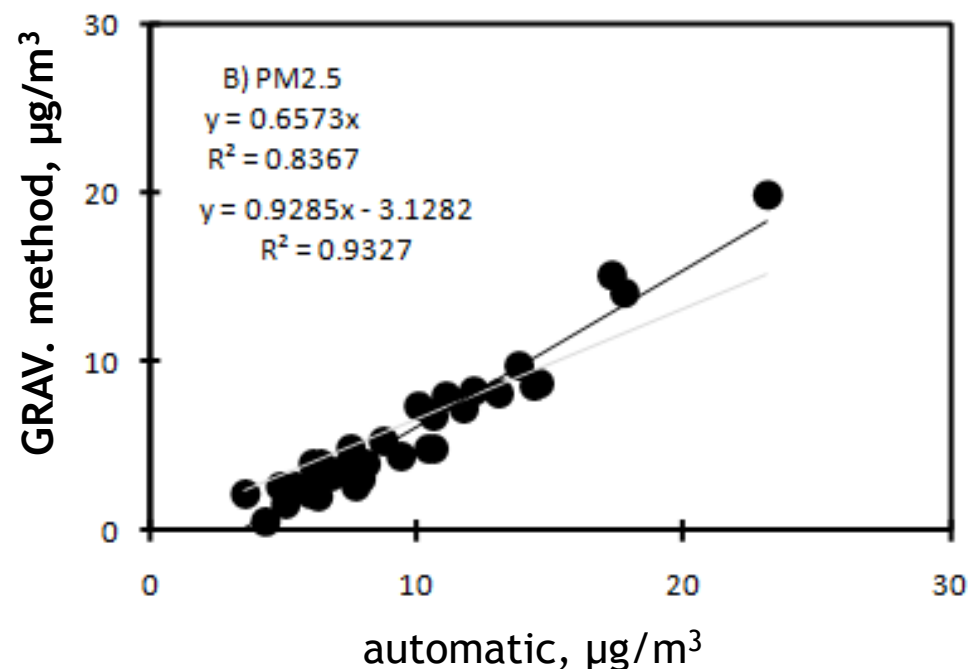
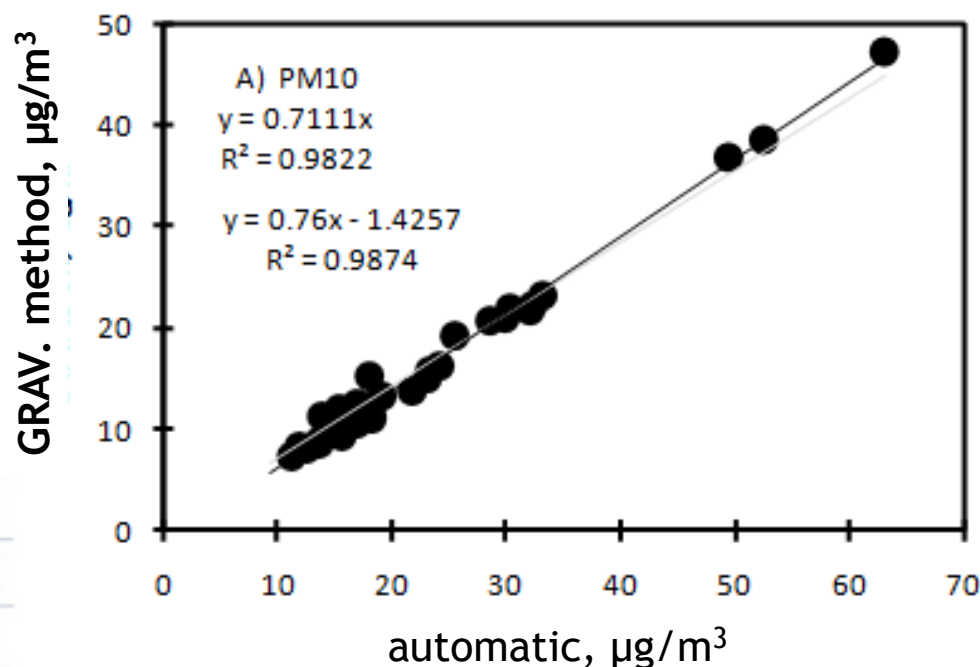


1. PM_{10} and $PM_{2.5}$ levels

- method-01: reference - manual gravimetry
- method-02: automatic

We recommend to convert PM_{10} and $PM_{2.5}$ data obtained with automatic instruments to gravimetric equivalent data.
For this intercomparisons are necessary

Validation of the automatic measurements Intercomparisons



Data evaluation:

automatic data are valid if they fit A or B:

A) $Y = a \cdot X$; $r^2 \geq 0.8$

B) $Y = a \cdot X + b$; $r^2 \geq 0.8$; $\text{abs}(b) < 5$

Y= gravimetric method,
X= Automatic analyzer

$\text{PM}_{10} (\text{grav}) = 0.71 \cdot \text{PM}_{10} (\text{automatic})$

$\text{PM}_{2.5} (\text{grav}) = 0.65 \cdot \text{PM}_{2.5} (\text{automatic})$



dust air quality



1. PM_{10} and $PM_{2.5}$ levels

-method-01: reference - manual gravimetry

-method-02: automatic

Manual gravimetry

automatic

advantage: reference method

high time resolution, 1h

disadvantage: poor time resolution, 24-h average
manual work
takes 3 days to know PM_{10}

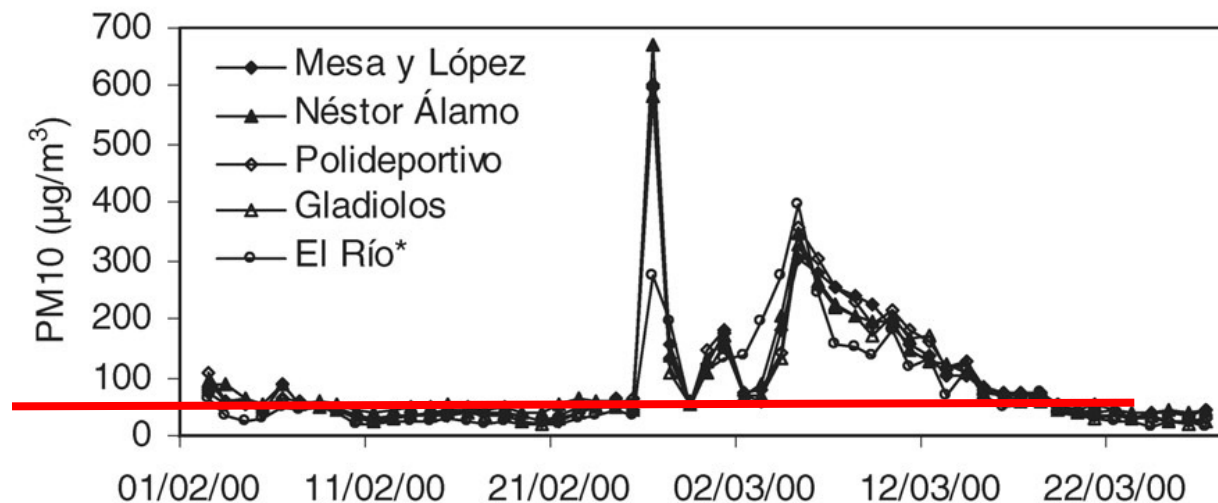
Needs validation

we recommend to use the two methods:

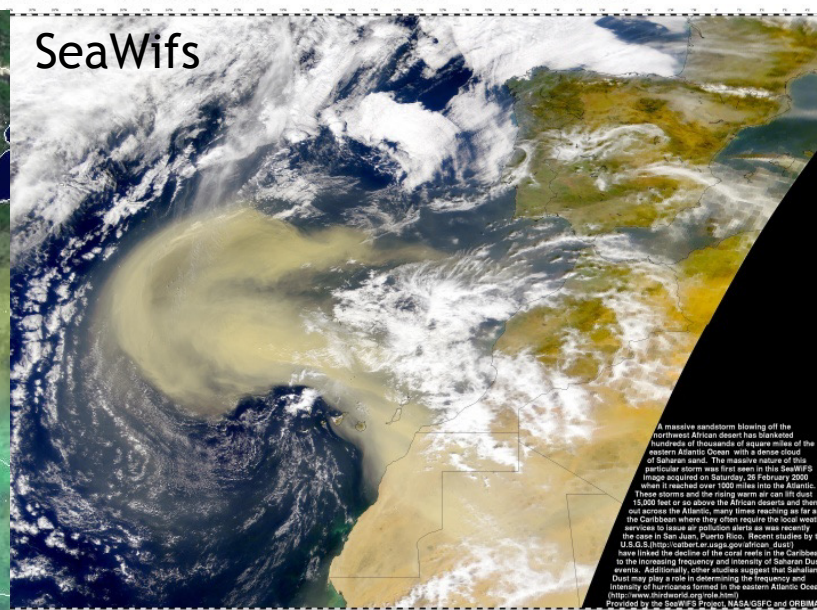
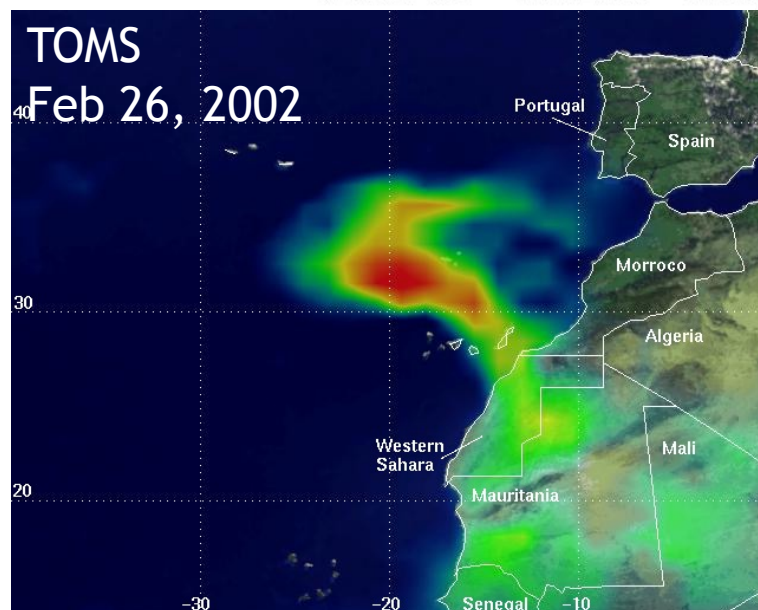
-automatic, continuously

-gravimetric: intercomparisons - 1 month summer, 1 month winter

Air quality stations at Tenerife Island



The WHO recommend PM_{10} (24-h) do not exceed $50 \mu\text{g}/\text{m}^3$



Viana et al., Atmospheric Environment, 2002

dust, aerosols and pollutants

in-situ observations

PM₁₀ and PM_{2.5} levels

PM₁₀ and PM_{2.5} composition

complementary observations

observation network



dust air quality



1. PM_{10} and $PM_{2.5}$ levels

-method-01: reference - manual gravimetry

-method-02: automatic

Manual gravimetry

advantage: reference method

CHEMICAL ANALYSIS

disadvantage: poor time resolution, 24-h average
manual work
takes 3 days to know PM_{10}

automatic

high time resolution, 1h

Needs validation

we recommend to use the two methods:

- automatic, continuously
- gravimetric: intercomprisons, 1 month summer, 1 month winter

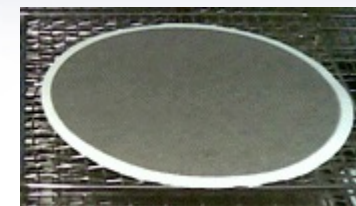
bulk chemical composition

PM samples: $\left\{ \begin{array}{l} \text{fine + coarse (TSP, PM}_{10}\text{)} \\ \text{fine (PM}_{2.5}\text{, PM}_1\text{)} \end{array} \right.$

Saharan dust



Urban particles



PM ($\mu\text{g}/\text{m}^3$) = **dust** + **trace elements** + **ions** (SO_4^- , NO_3^- , NH_4^+ , Na^+ , Cl^-) + OC + EC

Elemental Composition:

Major elements (Al, Si, Ca, K, Na, Mg) + trace elements (P, Li, Be, Sc, Ti, V, Cr, Mn, Co, Ni, Cu, Zn, Ga, Ge, As, Se, Rb, Sr, Y, Zr, Nb, Mo, Cd, Sn, Sb, Cs, Ba, La, Ce, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, Lu, Hf, Ta, W, Tl, Pb, Bi, Th, U)

Inductively coupled plasma
Atomic Emission Spectroscopy
ICP-AES

Inductively coupled plasma
Mass spectroscopy
IPC-MS

Destructive techniques

Ions: SO_4^- , NO_3^- , NH_4^+ , Na^+ , Cl^-

Ion Chromatography, ICP-AES, ICP-MS, selective electrodes and colorimetry

Destructive techniques

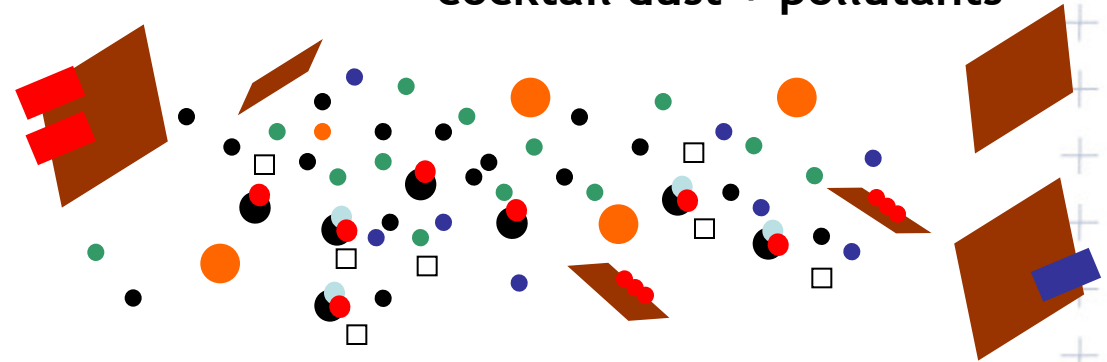
Thermal/optical reflectance (TOR) and/or thermal/optical transmission (TOT)

destructive techniques

XRF, PIXE, INAA : none destructive techniques



people live in cities and breath a cocktail dust + pollutants



PM_{10} : dust + sea salt + (sulphate + nitrate + organic matter + black carbon + metals) ..

$PM_{2.5}$: dust + sea salt + (sulphate + nitrate + organic matter + black carbon + metals) ..

PM_{10} : dust + sea salt + vehicle exhaust + oil refining + power plants + ships +...

$PM_{2.5}$: dust + sea salt + vehicle exhaust + oil refining + power plants + ships +...

dust, aerosols and pollutants

in-situ observations

PM₁₀ and PM_{2.5} levels

PM₁₀ and PM_{2.5} composition

complementary observations

let's build our observation network !!!

in-situ observations



dust air quality



in-situ observations

PM_{10} and $PM_{2.5}$ levels

PM_{10} and $PM_{2.5}$ composition

complementary observations

meteorology:

wind, temperature, relative humidity, pressure

gaseous pollutants (**reference methods**):

NO_x : vehicle exhausts, ships, oil refining, power plants..

SO_2 :, ships, oil refining, power plants

CO: vehicle exhausts



Examples of reference methods:

NO_x: chemiluminiscense. EN 14211: 2006

SO₂: fluorescense. EN 14212: 2006

CO: NDIR absorption. EN 14626: 2006

O₃: NDIR absorption. EN 14625: 2006



dust air quality



Recommended priorities

Level 1 (max priority) - PM_{10} and $PM_{2.5}$ levels - automatic methods

Level 1 (max priority) - meteorology (wind, T, RH, P, rain)

Level 2 - PM_{10} and $PM_{2.5}$ levels - complementary gravimetric method

Level 3 - gaseous pollutants: NO_x , SO_2 , CO,...

Level 4 - PM_{10} and $PM_{2.5}$ chemical composition

dust, aerosols and pollutants

in-situ observations

PM_{10} and $PM_{2.5}$ levels

PM_{10} and $PM_{2.5}$ composition

complementary observations

observation network



dust air quality



Recommended priorities

Level 1

- PM_{10} and $PM_{2.5}$ levels - automatic methods

Level 1

- meteorology (wind, T, RH, P, rain)

Level 2

- PM_{10} and $PM_{2.5}$ levels - complementary gravimetric method

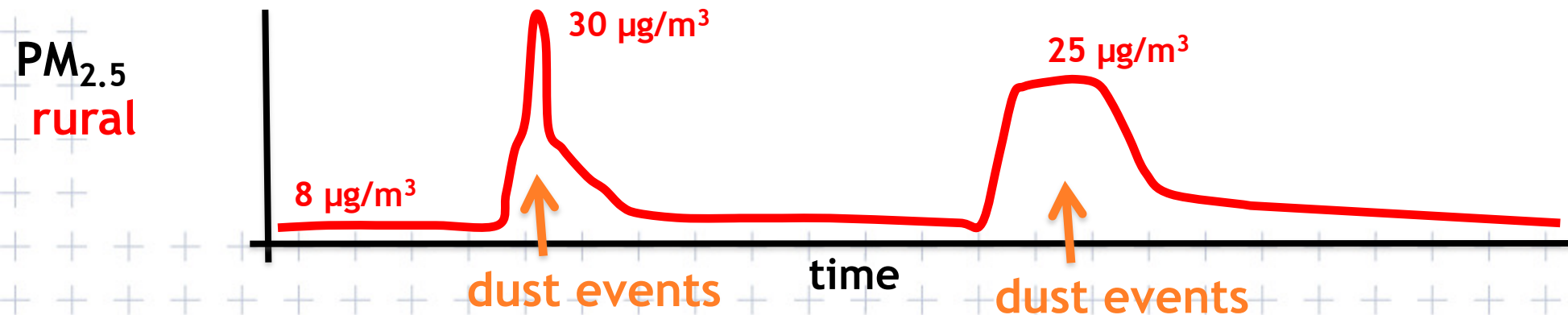
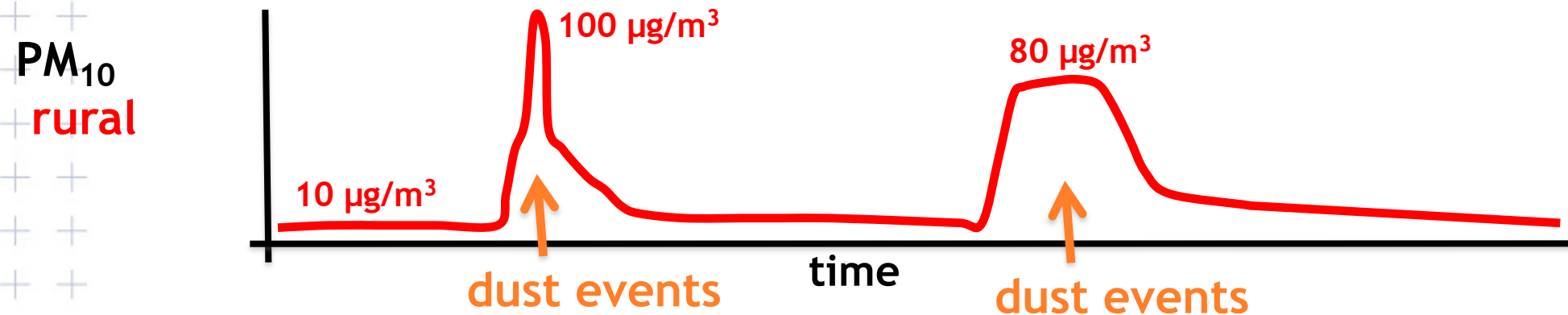
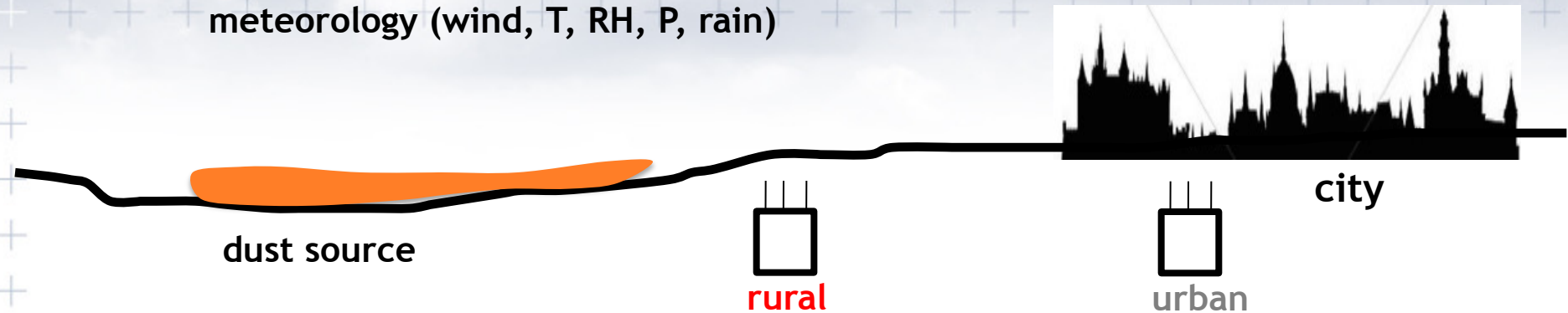
Level 3

- gaseous pollutants: NO_x , SO_2 , CO,...

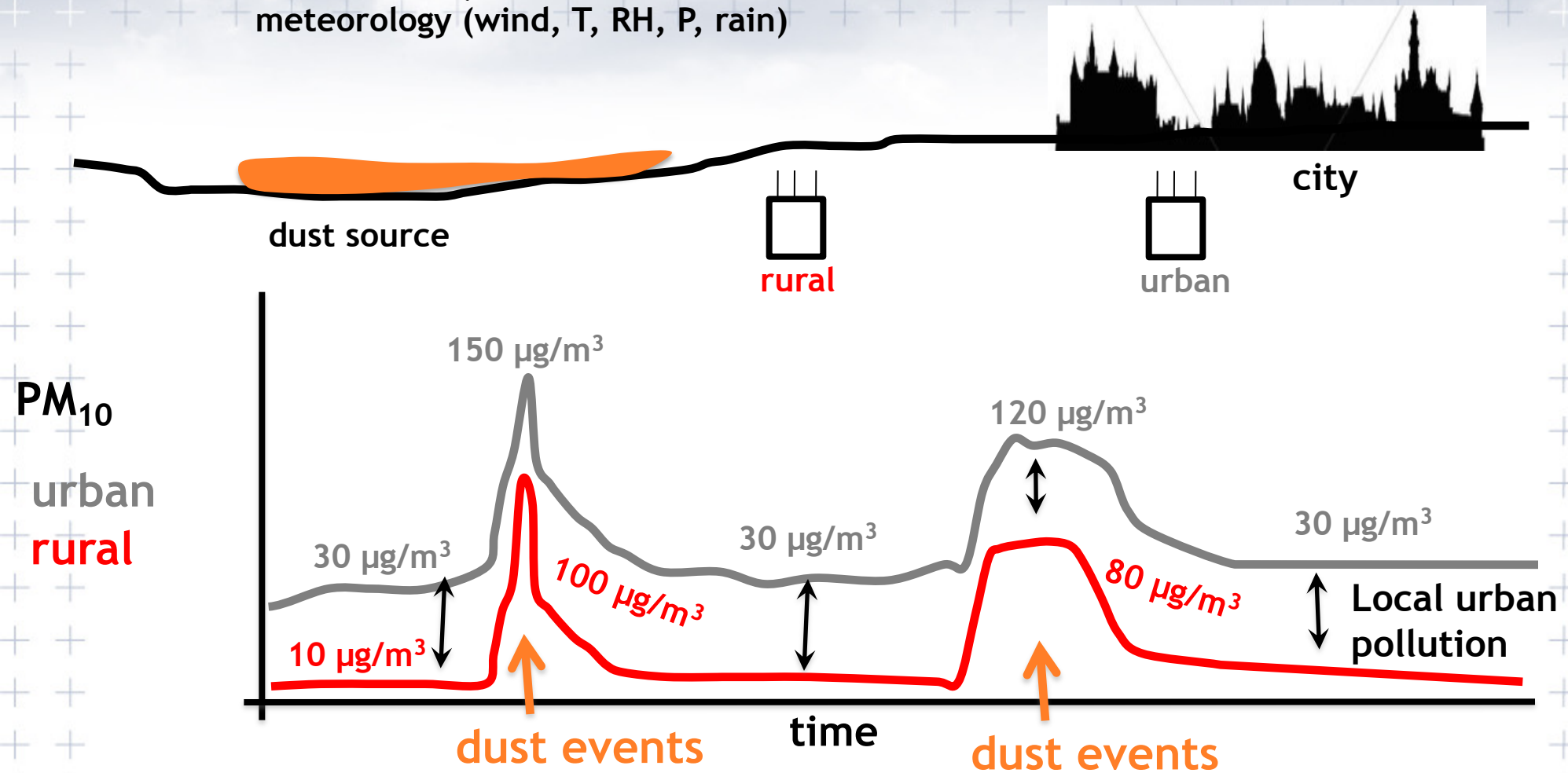
Level 4

- PM_{10} and $PM_{2.5}$ chemical composition

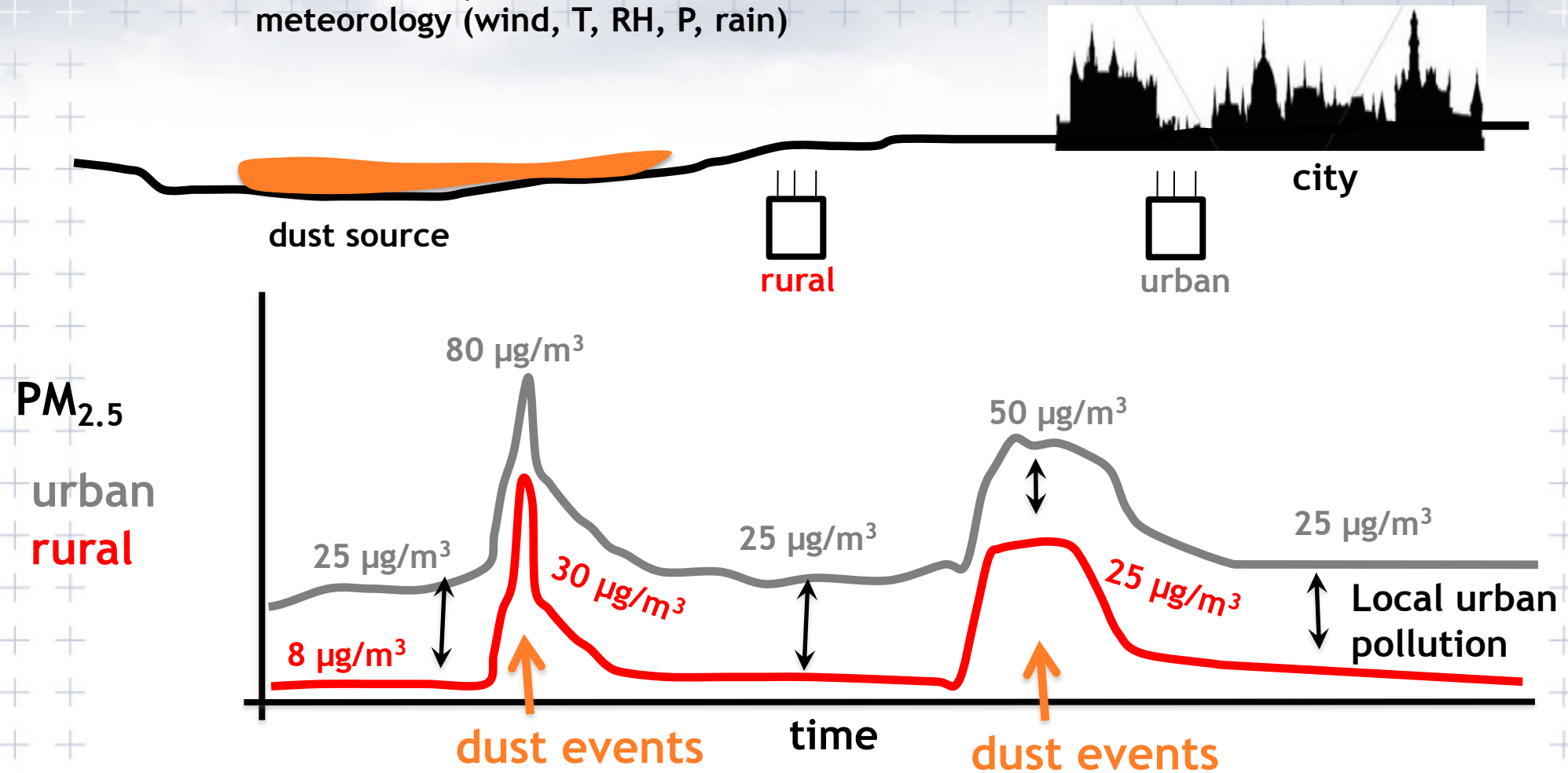
Level 1 PM_{10} and $PM_{2.5}$ - automatic methods meteorology (wind, T, RH, P, rain)



Level 1 PM_{10} and $PM_{2.5}$ - automatic methods meteorology (wind, T, RH, P, rain)



Level 1 PM_{10} and $PM_{2.5}$ - automatic methods meteorology (wind, T, RH, P, rain)





dust air quality



Recommended priorities

Level 1

- PM_{10} and $PM_{2.5}$ levels - automatic methods

Level 1

- meteorology (wind, T, RH, P, rain)

Level 2

- PM_{10} and $PM_{2.5}$ levels - complementary gravimetric method

Level 3

- gaseous pollutants: NO_x , SO_2 , CO,...

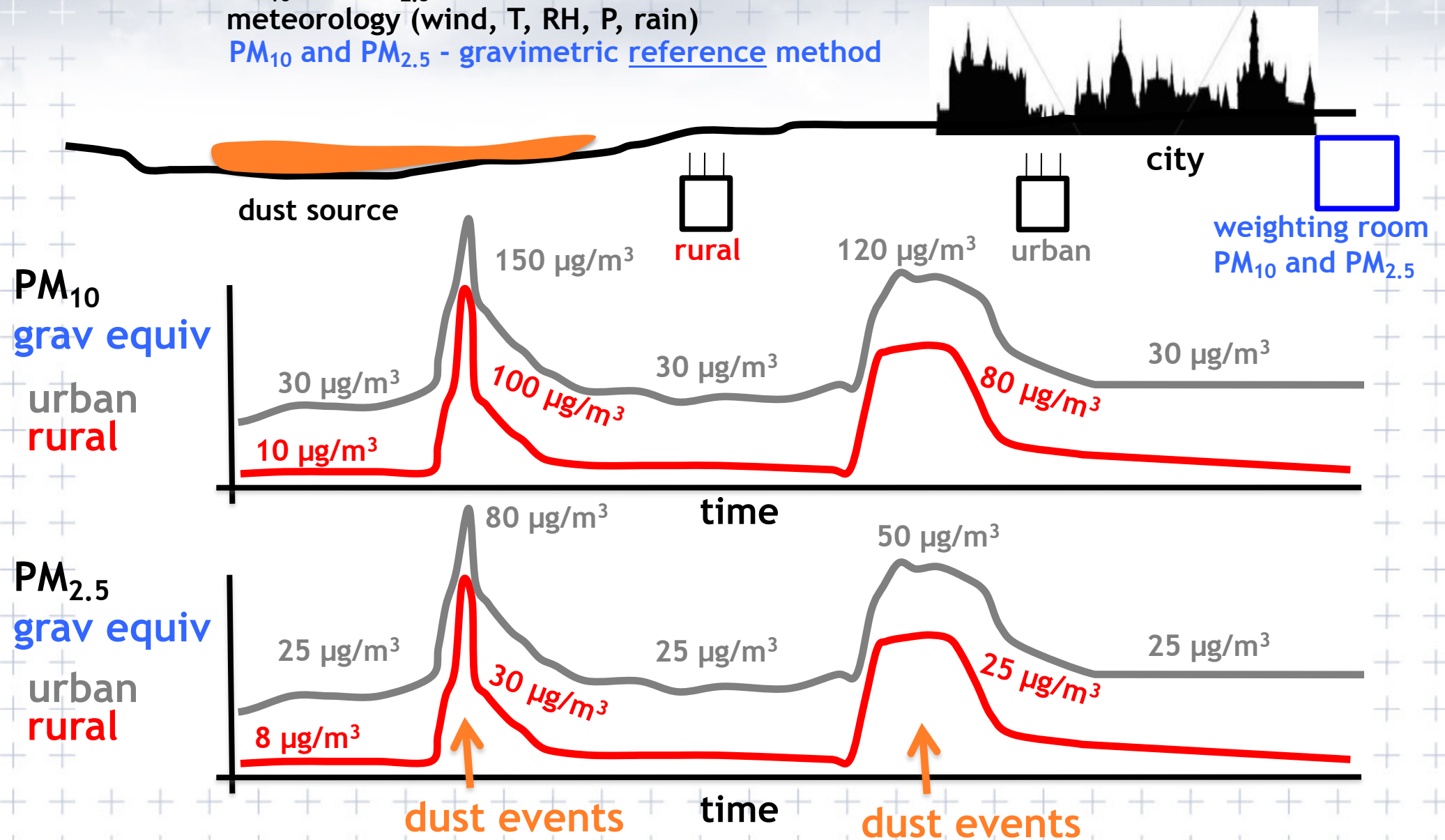
Level 4

- PM_{10} and $PM_{2.5}$ chemical composition

Level 2

PM₁₀ and PM_{2.5} - automatic methods
meteorology (wind, T, RH, P, rain)

PM₁₀ and PM_{2.5} - gravimetric reference method





dust air quality



Recommended priorities

Level 1 (max) - PM_{10} and $PM_{2.5}$ levels - automatic methods

Level 1 (max) - meteorology (wind, T, RH, P, rain)

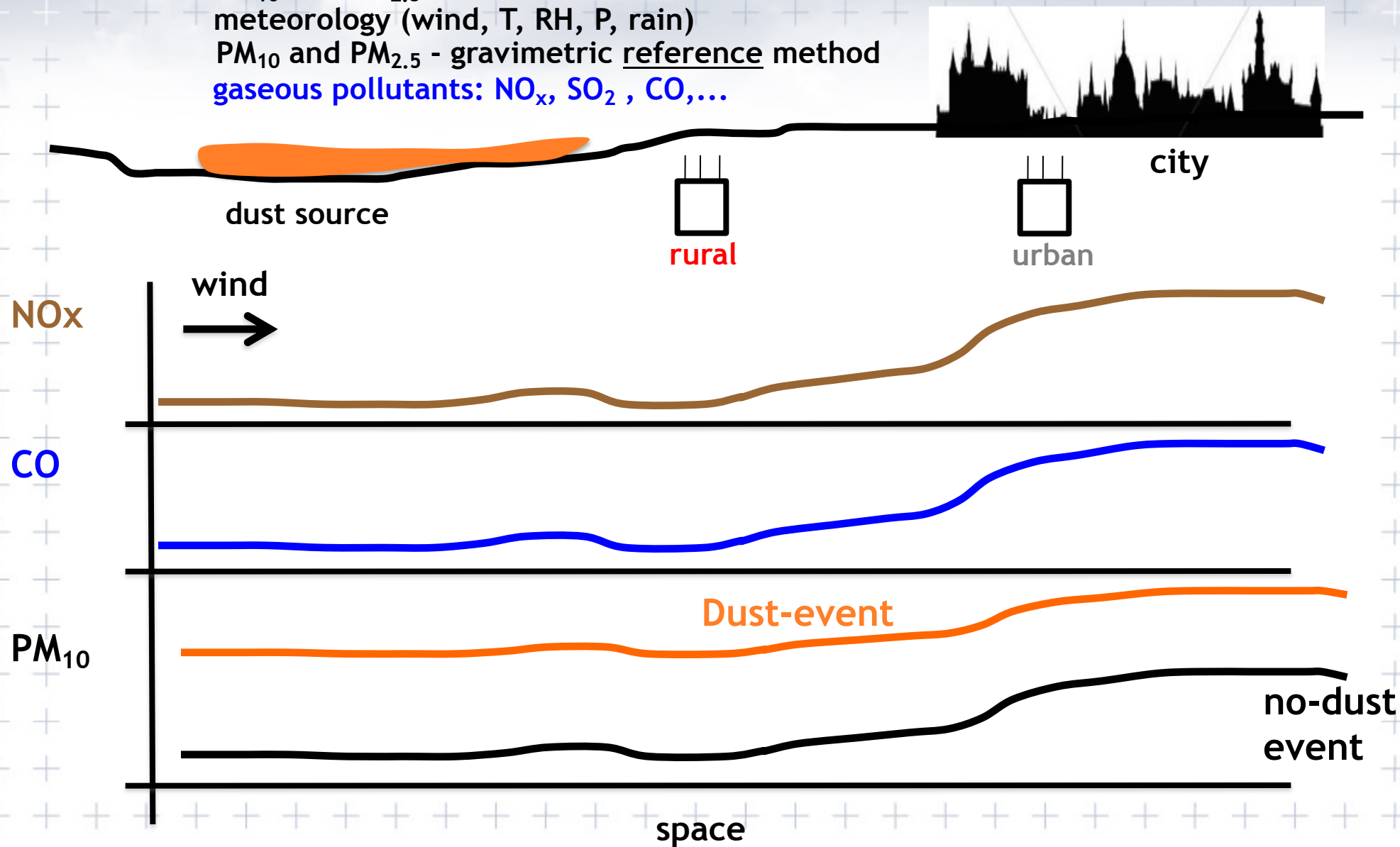
Level 2 - PM_{10} and $PM_{2.5}$ levels - complementary gravimetric method

Level 3 - gaseous pollutants: NO_x , SO_2 , CO ,...

Level 4 - PM_{10} and $PM_{2.5}$ chemical composition

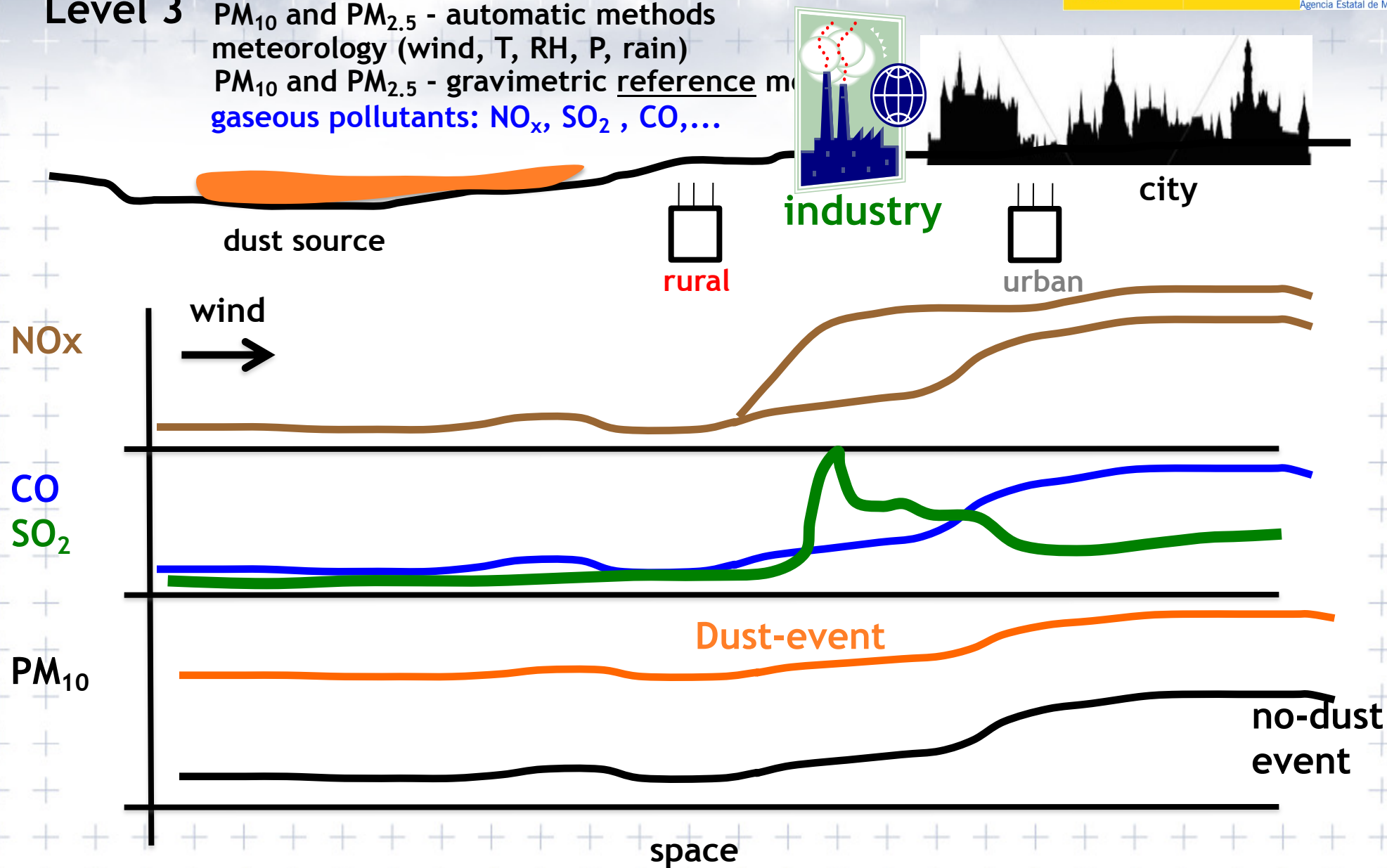
Level 3

PM₁₀ and PM_{2.5} - automatic methods
meteorology (wind, T, RH, P, rain)
PM₁₀ and PM_{2.5} - gravimetric reference method
gaseous pollutants: NO_x, SO₂, CO,...



Level 3

PM₁₀ and PM_{2.5} - automatic methods
meteorology (wind, T, RH, P, rain)
PM₁₀ and PM_{2.5} - gravimetric reference method
gaseous pollutants: NO_x, SO₂, CO,...





dust air quality



Recommended priorities

Level 1 (max) - PM_{10} and $PM_{2.5}$ levels - automatic methods

Level 1 (max) - meteorology (wind, T, RH, P, rain)

Level 2 - PM_{10} and $PM_{2.5}$ levels - complementary gravimetric method

Level 3 - gaseous pollutants: NO_x , SO_2 , CO,...

Level 4 - PM_{10} and $PM_{2.5}$ chemical composition

Level 4 PM_{10} and $PM_{2.5}$ - automatic methods
 meteorology (wind, T, RH, P, rain)
 PM_{10} and $PM_{2.5}$ - gravimetric reference methods
 gaseous pollutants: NO_x , SO_2 , CO,...
 PM_{10} and $PM_{2.5}$ - chemical composition

